

CHAPTER 21 STUDY GUIDE

Hydrocarbons

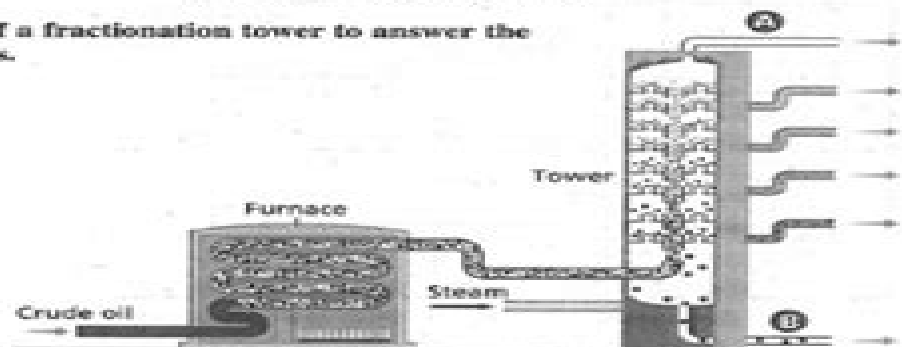
Section 21.1 Introduction to Hydrocarbons

In your textbook, read about natural sources of hydrocarbons and rating gasolines.

In the space at the left, write the word or phrase in parentheses that correctly completes the statement.

- _____ 1. (Petroleum, Natural gas) is a mixture of thousands of organic compounds.
- _____ 2. The boiling of petroleum and collection of its components is called (sedimentation, fractional distillation).
- _____ 3. In the process known as (cracking, knocking), heavier petroleum fractions are converted to gasoline by breaking their large molecules into smaller ones.
- _____ 4. A gasoline's ability to burn evenly and prevent knocking is expressed by its (hexane, octane) rating.

Use the diagram of a fractionation tower to answer the following questions.



5. How does the temperature inside the tower vary from bottom to top? _____
6. In what physical state is the material collected from pipe A? _____
7. What is the material collected from pipe A used for? _____
8. What is the material collected from pipe B used for? _____
9. Which pipe, A or B, collects hydrocarbons with higher boiling points? _____
10. Which pipe, A or B, collects smaller hydrocarbons? _____

Hydrocarbons Study Guide Answers

**Steven E. Lucking, Frank A.
Maffei, Robert F. Tamburro, Neal J.
Thomas**



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Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise

Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph

Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications *Hazmat Chemistry Study Guide*

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