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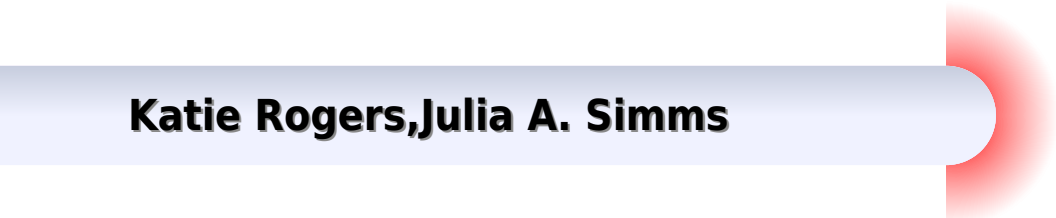
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**Katie Rogers,Julia A. Simms**



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web the graph of the vapor pressure of water versus temperature in figure pageindex 3 indicates that the vapor pressure of water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

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web nov 1 2003 droplet evaporation modeling by the distillation curve model accounting for kerosene fuel and elevated pressures

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web vapor pressure 0 1 psi at 100 f ntp 1992 vapor density relative to air 4 5 ntp 1992 heavier than air will sink specific gravity 0 8 at 59 f uscg 1999 less dense than water will float

[vapor pressure as a function of temperature the trend lines are](#) - Jun 04 2023

web in fact when the temperature increases from 20 to 60 c the vapor pressure of kerosene rises from 4 to 18 mbar 34 this suggests that kerosene cannot be entirely vaporized at 60 c

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web kerosene icsc 0663 physical chemical information physical state appearance low viscosity liquid with characteristic odour physical dangers as a result of flow agitation etc electrostatic charges can be generated vapour pressure kpa at 37 8 c 1 3 7 exposure health effects