

- Directions: (1) Substantiate your answers with formulas for finding using one or more of the following:
 (2) Balance each equation.
 (3) Identify the type reaction, products, decomposition, single replacement, double replacement, or combination.

1. Aluminum metal reacts with oxygen (air) to form aluminum oxide and oxygen diatomic molecules.

Given: _____

_____ \rightarrow _____ \rightarrow _____ \rightarrow _____

2. Iron(III)oxide reacts with carbon (C) to form

Given: _____

_____ \rightarrow _____ \rightarrow _____

3. Calciumphosphorus pentoxide can be produced from the elements calcium and phosphorus. Write down chemical phosphorus reactions that are put together. It is a combination.

Given: _____

_____ \rightarrow _____ \rightarrow _____

4. Ammoniahydrogen chloride is formed by the reaction of ammonia and hydrogen gas.

Given: _____

_____ \rightarrow _____ \rightarrow _____

5. Ammoniahydrogen chloride gas is produced when ammonia reacts (NH₃) hydrochloric acid to form solid iron (NH₄) chloride and water vapor.

Given: _____

_____ \rightarrow _____ \rightarrow _____

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