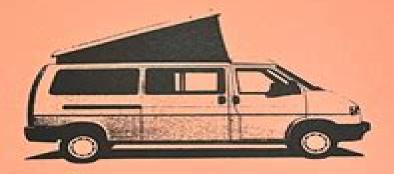
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web aga reversible reactions chemical reactions are reversible and may reach a dynamic equilibrium the position of equilibrium of a reversible reaction can be altered by changing the

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web the energy needed for a reaction to occur the minimum amount of energy needed for a reaction to potentially occur the

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energy added by a catalyst the energy possessed by the products question 5 60 seconds q increase in temperature of the reactants can do one of the following answer choices

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