May 2010

SEC Chemistry Paper 1

Marking Scheme

Section A

(a)		ectron -1					
(b			Electrons	Neutrons	Protons	Mass number	
	Λ	.tom R	17	18			
	Lie	on Q 2-			8	16	
	Te	on T *	2			7	
	A	tom W	17			37	56.2
	(iii)	Ranc	i W			both correct if one correct	
(a)		Moles Moles Ratio	carbon = 85. hydrogen = 1 1C: 2H E.F.		Award 0 i		6 mar
(a)		Moles Ratio If E.F	carbon = 85. hydrogen = 1 1C: 2H E.F.	4.29 / 1 = 14.2	Award 0 i	f one correct	6 mar
(a)	(ii)	Moles Moles Ratio If E.F. M.F. =	s carbon = 85. hydrogen = 1 IC : 2H E.F. is given as (C = 3 x EF = C ₃ H ₆	(4.29 / 1 = 14.2 = CH ₂ TH ₂) ₀ deduct ½ CHBrCH ₂ Br	Award 0 i	cept C ₃ H ₆ Br ₂	6 mar
	(ii)	Moles Moles Ratio If E.F. M.F. = E.F. = C ₃ H ₆	s carbon = 85. s hydrogen = 1 IC : 2H E.F. is given as (C = 3 x EF C ₃ H ₆ + Br ₂ © CH ₃ 6	(4.29 / 1 = 14.2 = CH ₂ TH ₂) ₀ deduct ½ CHBrCH ₂ Br	Award 0 i	cept C ₃ H _a Br ₂ ulae award 2 award 1 s orange-brown	6 mar
	(ii) (iii)	Moles Moles Ratio If E.F. M.F. = C ₃ H ₆ Red-o	s carbon = 85. s hydrogen = 1 1C : 2H E.F. is given as (C = 3 x EF = C ₃ H ₆ + Br ₂ © CH ₃ G	(4.29 / 1 = 14.2 = CH ₂ "H ₂) ₀ deduct ½ CHBrCH ₂ Br All If no	Award 0 i	cept C ₃ H ₆ Br ₂ alae award 2 award 1 s orange-brown	

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Sebastian Brünink

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- 1. Understanding the eBook Mark Scheme May Chemistry 2013 Paper 1c
 - The Rise of Digital Reading Mark Scheme May Chemistry 2013 Paper 1c
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Mark Scheme May Chemistry 2013 Paper 1c
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mark Scheme May Chemistry 2013 Paper 1c
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mark Scheme May Chemistry 2013 Paper 1c
 - Personalized Recommendations
 - Mark Scheme May Chemistry 2013 Paper 1c User Reviews and Ratings
 - Mark Scheme May Chemistry 2013 Paper 1c and Bestseller Lists

- 5. Accessing Mark Scheme May Chemistry 2013 Paper 1c Free and Paid eBooks
 - Mark Scheme May Chemistry 2013 Paper 1c Public Domain eBooks
 - Mark Scheme May Chemistry 2013 Paper 1c eBook Subscription Services
 - Mark Scheme May Chemistry 2013 Paper 1c Budget-Friendly Options
- 6. Navigating Mark Scheme May Chemistry 2013 Paper 1c eBook Formats
 - o ePub, PDF, MOBI, and More
 - Mark Scheme May Chemistry 2013 Paper 1c Compatibility with Devices
 - Mark Scheme May Chemistry 2013 Paper 1c Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mark Scheme May Chemistry 2013 Paper 1c
 - Highlighting and Note-Taking Mark Scheme May Chemistry 2013 Paper 1c
 - Interactive Elements Mark Scheme May Chemistry 2013 Paper 1c
- 8. Staying Engaged with Mark Scheme May Chemistry 2013 Paper 1c
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mark Scheme May Chemistry 2013 Paper 1c
- 9. Balancing eBooks and Physical Books Mark Scheme May Chemistry 2013 Paper 1c
 - \circ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mark Scheme May Chemistry 2013 Paper 1c
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mark Scheme May Chemistry 2013 Paper 1c
 - Setting Reading Goals Mark Scheme May Chemistry 2013 Paper 1c
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mark Scheme May Chemistry 2013 Paper 1c
 - Fact-Checking eBook Content of Mark Scheme May Chemistry 2013 Paper 1c
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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