

Very LOW concentrations : ppm, ppb & ppt

Certain gases, such as the greenhouse gases (for example, carbon dioxide, water vapour, methane, and ozone), occur in the atmosphere in miniscule amounts. In a random air sample from the troposphere, for example, you would likely find only about 350 molecules of carbon dioxide for every one million molecules of air mixture. Scientists would express this amount as 350 parts per million (ppm).

Gas concentrations can also be expressed in mass units. With gases in the atmosphere, we usually think in terms of volume and may express this as parts per million by volume (ppmv). Some substances occur in such small amounts that scientists measure them in even smaller amounts such as parts per billion by volume (ppbv) or even parts per trillion by volume (pptv).

Part per million:

- 1 can of coke in a 30-metre swimming pool of water
- 1 x 100g chocolate bar on a football pitch

Part per billion:

- 1 bad apple/1,000,000 barrels of apples
- 1 pinch of salt in 10 tons of crisps

Part per trillion:

- 1 flea/360,000,000 elephants

Parts-per million (ppm)	Coefficient / Ratio	Percent (%)	Parts per billion (ppb)	Parts per trillion (ppt)
1 ppm	1×10^{-6}	0.0001%	1000 ppb	1×10^3 ppt
10 ppm	1×10^{-5}	0.001%	10000 ppb	1×10^4 ppt
100 ppm	1×10^{-4}	0.01%	100000 ppb	1×10^5 ppt
1000 ppm	0.001	0.1%	1×10^5 ppb	1×10^6 ppt
10,000 ppm	0.010	1%	1×10^7 ppb	1×10^8 ppt
100000 ppm	0.100	10%	1×10^8 ppb	1×10^{11} ppt
1,000,000 ppm	1.000	100%	1×10^9 ppb	1×10^{12} ppt

F332 June 2013 Paper

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Table of Contents F332 June 2013 Paper

1. Understanding the eBook F332 June 2013 Paper
 - The Rise of Digital Reading F332 June 2013 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying F332 June 2013 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F332 June 2013 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from F332 June 2013 Paper
 - Personalized Recommendations
 - F332 June 2013 Paper User Reviews and Ratings
 - F332 June 2013 Paper and Bestseller Lists
5. Accessing F332 June 2013 Paper Free and Paid eBooks
 - F332 June 2013 Paper Public Domain eBooks
 - F332 June 2013 Paper eBook Subscription Services
 - F332 June 2013 Paper Budget-Friendly Options

6. Navigating F332 June 2013 Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - F332 June 2013 Paper Compatibility with Devices
 - F332 June 2013 Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of F332 June 2013 Paper
 - Highlighting and Note-Taking F332 June 2013 Paper
 - Interactive Elements F332 June 2013 Paper
8. Staying Engaged with F332 June 2013 Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F332 June 2013 Paper
9. Balancing eBooks and Physical Books F332 June 2013 Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F332 June 2013 Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine F332 June 2013 Paper
 - Setting Reading Goals F332 June 2013 Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of F332 June 2013 Paper
 - Fact-Checking eBook Content of F332 June 2013 Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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