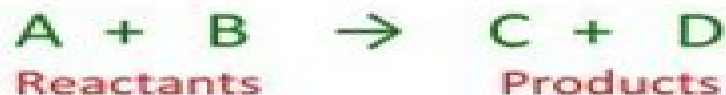


Chapter 7: Chemical Reactions

A chemical reaction: a process in which at least one new substance is formed as the result of a chemical change.



Evidence that a chemical reaction has occurred:

- Color change
- Formation of a solid (in a previously clear solution)
- Formation of a gas
- Emission of light
- Emission or absorption of heat

A chemical equation: a written statement that uses **symbols** and **formulas** instead of words to describe the changes that occur in a chemical reaction


Ex.



Special symbols are often used in chemical equations to express the state of reactants and products

(s) = solid
(l) = liquid
(g) = gas
(aq) = aqueous

substance dissolved in H₂O



Ex.



Mcgraw Chemical Equations Study Guide

Wolfgang Guggemos



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