Exam-style questions and sample answers have been written by the authors. In examinations, the way marks are awarded may be different.

> Coursebook answers

Chapter 1

Science in context: Lord of the rings

- Jupiter and Saturn are the biggest of the planets in our solar system and are made of hydrogen and helium in the gaseous state. There is a progression outwards from the Sun in terms of the structure of the planets. Those nearest the Sun are rocky and relatively small; beyond Mars the planets are gaseous.
- 2 Hydrogen molecules are the smallest of all the elements and therefore the forces between them are very small. The gas needs to be highly compressed (so that the molecules are close together) and at a very low temperature (so that the molecules are moving slowly) in order for it to condense into a liquid.

Questions

- 1 a freezing
 - b boiling
 - condensation
- 2 a methane
 - ethanol and mercury; they have melting points below room temperature, but boiling points above room temperature
 - the impurity lowers the freezing point of the liquid
- 3 a a volatile liquid is one that evaporates easily; it has a low boiling point
 - b ethanol > water > ethanoic acid; ethanol is the most volatile, ethanoic acid is the least volatile
 - B; both the melting point and boiling point are above room temperature; therefore, it is a solid

- 4 a if heated strongly then the temperature rises very quickly, the melting and boiling points are close together, the liquid stage is not seen as it boils quickly
 - b you would need to heat the solid slowly so that the temperature rise is not too quick; you could use an electrical heater so you can control temperature more carefully; you could use an oil bath to heat the solid and carefully control the temperature so it is kept between the melting point (114 °C) and boiling point (184 °C) of iodine

Experimental skills 1.1

- Substance B is a pure substance as the curve is flat while the substance melts. Substance A is a mixture as there is no single melting point – that region of the curve is sloped.
- Take temperature readings at shorter time intervals (e.g. every 30s); use of a digital thermometer would give more accurate readings / use of a temperature sensor linked to a computer would enable the readings to be taken continuously and enable the graph to be plotted as the readings were being taken. The samples can be re-heated and cooled again, allowing duplicate sets of data to be collected.

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L Darling-Hammond

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