

A 23.8 L balloon is filled with 88g CO_2
at 15°C , what is the pressure in

kPa?

$$PV = nRT$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$PV = nRT$$

$$P(23.8\text{L}) = (2)$$

$$V = \underline{23.8\text{L}}$$

$$T = 15^\circ\text{C} + 273 = \underline{288\text{K}}$$

$$P = ? \text{ kPa}$$

$$88\text{g CO}_2 \times \frac{1\text{mol CO}_2}{(12 + 16 \times 2)} = \underline{2\text{mol}}$$

44g

Ideal Gas Law Problems And Solutions Atm

James L. Flowers, Princeton
Review, Theodore Silver



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Thermodynamics Problem Solving in Physical Chemistry Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20 30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Provides instructor access to a visual map depicting how all equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics questions for major tests

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Ruskell,2008 **Environmental Calculations** Robert G. Kunz,2011-12-14 Consolidates information and technical calculations for a wide variety of environmental factors Operating a business facility of any size especially a manufacturing location requires environmental permits from a number of governmental regulatory agencies responsible for protecting human health and the environment Environmental Calculations A Multimedia Approach provides an essential one stop reference for the necessary technical calculations to obtain a broad range of such permits Along with clear concise and factual explanations the text also includes relevant equations examples and case studies to support and clarify the calculations Filled with the rich experience from the author s years of work in environmental permitting the coverage features An introduction to the major concepts and practice in the permitting process Key concepts in environmental chemistry such as the ideal gas law vapor pressure reaction stoichiometry and heat effects Air pollution control Water wastewater Solid hazardous waste Noise generation propagation and control Radiation radioactive decay An all around guide for environmental permitting in many contexts Environmental Calculations A Multimedia Approach is a must have for anybody concerned with environmental assessment and compliance as well as those reviewing issuing and monitoring environmental permits **U Can: Chemistry I For Dummies** John T. Moore,Chris Hren,Peter J. Mikulecky,2015-07-21

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Ideal Gas Law Problems And Solutions Atm Book Review: Unveiling the Power of Words

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