

Key

Ideal Gas Law Worksheet $PV = nRT$

Use the ideal gas law, "PerV-nRT", and the universal gas constant $R = 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}}$ to solve the following problems:

If pressure is needed in kPa then convert by multiplying by $101.3 \text{ kPa} / 1 \text{ atm}$ to get
 $R = 8.31 \text{ kPa} \cdot \text{L} / (\text{K} \cdot \text{mole})$

- 1) If I have 4 moles of a gas at a pressure of 5.6 atm and a volume of 12 liters, what is the temperature?

$$PV = nRT$$

$$T = \frac{PV}{nR} = \frac{(5.6 \text{ atm})(12 \text{ L})}{4 \text{ mol} \cdot 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}}}$$

$$T = 204.63 \text{ K}$$

- 2) If I have an unknown quantity of gas at a pressure of 1.2 atm, a volume of 31 liters, and a temperature of 87°C , how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{1.2 \text{ atm} \cdot 31 \text{ L}}{0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}} \cdot 360 \text{ K}}$$

$$n = 1.2586 \text{ mol}$$

- 3) If I contain 3 moles of gas in a container with a volume of 60 liters and at a temperature of 400 K, what is the pressure inside the container?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{3 \text{ mol} \cdot 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}} \cdot 400 \text{ K}}{60 \text{ L}}$$

$$P = 1.642 \text{ atm}$$

$$\text{or } P = 166.29 \text{ kPa}$$

- 4) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of 56°C , what is the volume of the container that the gas is in?

$$PV = nRT$$

$$V = \frac{nRT}{P} = \frac{7.7 \text{ mol} \cdot 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}} \cdot 327 \text{ K}}{0.09 \text{ atm}}$$

$$V = 2310.93 \text{ L}$$

- 5) If I have 17 moles of gas at a temperature of 67°C , and a volume of 88.89 liters, what is the pressure of the gas?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{17 \text{ mol} \cdot 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}} \cdot 390 \text{ K}}{88.89 \text{ L}}$$

$$P = 5.34 \text{ atm}$$

$$\text{or } P = 540.61 \text{ kPa}$$

- 6) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{0.5 \text{ atm} \cdot 25 \text{ L}}{0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}} \cdot 300 \text{ K}}$$

$$n = 0.5075 \text{ mol}$$

Ideal Gas Additional Problems Holt

John Elliston Allen, Joan Bruce



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Holt Chemistry File, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

Computational Techniques and Applications, CTAC-83 John Noye, C. A. J. Fletcher, 1984 **Idealization IX** Niall Shanks, 1998 Here is presented for the first time a comprehensive review and analysis of the several roles played by idealization procedures in the logic mathematics and models that lie at the heart of modern twentieth century physics It is only through idealization of one form or another that the objects and processes of modern physics become tractable The essays in this volume will be of interest to all those who are concerned with the uses of models in physics and the relationships between models and the real world The essays in this volume cover the role of idealization in all the main areas of modern physics ranging from quantum theory relativity theory and cosmology to chaos theory Mathematics of Large Eddy Simulation of Turbulent Flows Luigi Carlo Berselli, Traian Iliescu, William J. Layton, 2006 The LES method is rapidly developing in many practical applications in engineering The mathematical background is presented here for the first time in book form by one of the leaders in the field **Difference Methods for Initial-Boundary-Value Problems and Flow Around Bodies** You-lan Zhu, Xi-chang Zhong, Bing-mu Chen, Zuo-min Zhang, 2013-06-29 Since the appearance of computers numerical methods for discontinuous solutions of quasi linear hyperbolic systems of partial differential equations have been among the most important research subjects in numerical analysis The authors have developed a new difference method named the singularity separating method for quasi linear hyperbolic systems of partial differential equations Its most important feature is that it possesses a high accuracy even for problems with singularities such as shocks contact discontinuities rarefaction waves and detonations Besides the thorough description of the method itself its mathematical foundation stability convergence theory of difference schemes for initial boundary value hyperbolic problems and its application to supersonic flow around bodies are discussed Further the method of lines and its application to blunt body problems and conical flow problems are described in detail This book should soon be an important working basis for both graduate students and researchers in the field of partial differential equations as well as in mathematical physics **Issues of the Day** Ian W.H. Parry, Felicia Day, 2010-09-23 Issues of the Day provides an easy way for students academics journalists policymakers and the public to learn about a diverse range of policy issues affecting the environment energy transportation and public health Each commentary gives a short assessment of a topic summarizing in a non technical way the current state of analysis or evidence on the issue along with selected recommendations for further reading The essays are written by world renowned scholars mostly economists and provide useful insights on policy problems that are often complex and poorly understood Some of the topics covered include air pollution hazardous waste voluntary environmental programs domestic U

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Automotive Industries, 1917 Automotive Industries, the Automobile, 1917 **Computational Techniques for Fluid Dynamics 1** Clive A.J. Fletcher, 2012-12-06. This well known 2 volume textbook provides senior undergraduate and postgraduate engineers, scientists and applied mathematicians with the specific techniques and the framework to develop skills in using the techniques in the various branches of computational fluid dynamics. A solutions manual to the exercises is in preparation. *The Automobile*, 1917 *The Future of Aeronautics* John Elliston Allen, Joan Bruce, 1971

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