

Concept Review

Section: What Is Matter?

1. Classify the following as a homogeneous, *O*, or a heterogeneous, *E*, mixture.

_____ a. a pail of sand and water

_____ b. air

_____ c. human blood

_____ d. banana split

_____ e. chocolate syrup

_____ f. sea water

2. Compare and contrast atoms and molecules.

3. Explain the difference between a pure substance and a homogeneous mixture.

4. Classify each of the following as an element or a compound.

_____ a. benzene, C_6H_6

_____ b. aluminum, Al

_____ c. aspirin, $C_9H_8O_4$

_____ d. titanium, Ti

_____ e. acetylene, C_2H_2

_____ f. zinc, Zn

5. Explain why elements and compounds are pure substances.

6. Determine which of the following are pure substances and which are mixtures.

_____ a. salt water

_____ b. isopropyl alcohol, C_3H_8O

_____ c. mercury, Hg

_____ d. ammonia, NH_3

_____ e. an egg yolk

_____ f. honey

Holt Chemistry Section 2 Answers

Robert R. Redfield



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