

- What are molecules? Give examples.
- What is a physical change and list examples of physical changes.
- What is a chemical change and give examples of chemical changes.

2.2 Main Body (Lesson presentation)

Educator starts lesson off with an exciting demonstration as mentioned in the introduction and explains the relevant concepts in terms of a chemical change. (20 min.)

- Matter is all around us and it undergoes changes all the time. These changes can be classified as **PHYSICAL CHANGES** OR **CHEMICAL CHANGES**.
- The **CONCEPTS** of **PHYSICAL CHANGE** and **CHEMICAL CHANGE** is shown in the table below. Educator engages learners in a discussion of the different aspects of physical and chemical change as mentioned in the table to re-enforce the concepts.

physical changes	chemical changes
common signs that a chemical change has occurred are: 1. Production of gas bubbles 2. Change in the way something melts 3. A release of energy such as a flash or a sound (like a firecracker) 4. A precipitate forms (two liquids mixed together form a solid and a liquid)	common signs that a physical change has occurred are: 1. change in the size 2. change in shape 3. change in colour, or 4. Change in state / phase of matter of a substance. 5. No new substance is produced.
Examples of physical changes • A grape when stepped on (changes shape) - Blowing up a balloon (changes size and shape) • Liquid water turning to ice (changes state of matter) • Liquid water turning to steam (changes state of matter) • Mixing salt and sugar (changes the appearance, but you can still separate the mixture) • Mixing water and salt (changes the appearance, but you can still separate the mixture)	Examples of chemical changes • Metal rusting (new substance formed) • Stomach digesting food (break down of food to new substances) • Plant carrying out photosynthesis (pulling water and carbon dioxide together to make sugar) • Mixing baking soda and vinegar (makes a neutral liquid and a gas)

- Educator discusses the results of the experiment with the learners.
- The liquid hydrogen peroxide decomposes (breaks up) to form oxygen gas and liquid water (the atomic model the using ball and stick as well as space filling diagrams to show the reactants and products in this reaction).
- The **MANGANESE DIOXIDE** is a catalyst and speeds up the reaction. A **CATALYST** is a chemical substance that changes the speed of a reaction without getting used up in the process. at the end of the reaction the catalyst can be recovered. Eg. **ENZYMES** in our body help to speed up the digestive process.

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Table of Contents Grade Caps Physical Sciences Lesson Plan

1. Understanding the eBook Grade Caps Physical Sciences Lesson Plan
 - The Rise of Digital Reading Grade Caps Physical Sciences Lesson Plan
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Caps Physical Sciences Lesson Plan
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Caps Physical Sciences Lesson Plan
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Caps Physical Sciences Lesson Plan
 - Personalized Recommendations
 - Grade Caps Physical Sciences Lesson Plan User Reviews and Ratings
 - Grade Caps Physical Sciences Lesson Plan and Bestseller Lists
5. Accessing Grade Caps Physical Sciences Lesson Plan Free and Paid eBooks
 - Grade Caps Physical Sciences Lesson Plan Public Domain eBooks
 - Grade Caps Physical Sciences Lesson Plan eBook Subscription Services

- Grade Caps Physical Sciences Lesson Plan Budget-Friendly Options
- 6. Navigating Grade Caps Physical Sciences Lesson Plan eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade Caps Physical Sciences Lesson Plan Compatibility with Devices
 - Grade Caps Physical Sciences Lesson Plan Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Caps Physical Sciences Lesson Plan
 - Highlighting and Note-Taking Grade Caps Physical Sciences Lesson Plan
 - Interactive Elements Grade Caps Physical Sciences Lesson Plan
- 8. Staying Engaged with Grade Caps Physical Sciences Lesson Plan
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade Caps Physical Sciences Lesson Plan
- 9. Balancing eBooks and Physical Books Grade Caps Physical Sciences Lesson Plan
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Caps Physical Sciences Lesson Plan
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Caps Physical Sciences Lesson Plan
 - Setting Reading Goals Grade Caps Physical Sciences Lesson Plan
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Caps Physical Sciences Lesson Plan
 - Fact-Checking eBook Content of Grade Caps Physical Sciences Lesson Plan
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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