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Purity Difference between compounds and mixtures Homogenous mixtures Heterogeneous mixtures Separation Techniques Decantation Filtration Sublimation Chromatography Distillation Fractional distillation Crystallization Centrifugation Exam Style Questions Chapter 3 Structure of Atom Atoms Elements Discovery of sub atomic particles Models of Atom Structure and Stability Atom and Ion Isotopes Radio Isotopes Electronic Arrangement Exam style Questions Chapter 5 Stoichiometry Elements Compounds Chemical Formula Word Equation Symbol Equation Balancing Equation Relative Atomic Mass Naming Compound Information from a chemical equation Definition of Mole The mole concept Molar Mass Important Formula Limiting reagent Reacting Masses Reacting masses and ratios Molar Volumes Concentration of Solutions Water of Crystallization Empirical and Molecular Formula Percentage Yield Percentage Purity Solved Examples of all the concepts Practice Questions Exam style Questions Chapter 6 Chemical Bonding Chemical Bond Ionic Bond Covalent Bond Metallic Bond Coordinate Bond Giant Structures Formula of positive and negative ions Exam style Questions Edexcel Certificate/International GCSE Chemistry Exam Practice Workbook (with Answers & Online Edition) CGP Books,2013-09-04

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