

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পরীক্ষার প্রয়োজক

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

চলকের ভর, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে উল্লিখিত N_2 -সংশ্লিষ্টে 10 mg এর মধ্যে কয় পরিমাণ চূর্ণ ও V_1 -সংশ্লিষ্টে 0.2 g 2% $\left(\frac{W}{V}\right)$ HCl চূর্ণের মধ্যে কয় পরিমাণ চূর্ণ।

যদিও N_2 -সংশ্লিষ্টে হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V_1 -সংশ্লিষ্টে হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায়শই রাসায়নিক উপাদান সন্নিবেশন ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।

$(CH_3NH_2 + H_2O \rightarrow CH_3NH_3^+ + OH^-)$ ফলে, কোনো কোনো ক্ষেত্রে কম কৃষিকর রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

চিত্র-১০



১. উদাহরণে চূর্ণের কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের চূর্ণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চূর্ণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিমিত্ত কীমের উপর রাইডার স্থাপন করলে কীমের প্রতি দৃষ্টান্তের জন্য যে চূর্ণ পাওয়া যায়, তাকে রাইডার চূর্ণক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায়শই রাসায়নিক উপাদান সন্নিবেশন ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৩. উদাহরণের চূর্ণকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M চূর্ণ। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M চূর্ণ হলে পাত্র H_2SO_4 চূর্ণ। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আমদান	ক্ষয়কারী রাসায়নিক উপাদান কয়েক সে মিনিট রাসায়নিক দ্রব্যকে চূর্ণের দ্বারা সংস্পর্শে তুলে ও অন্যান্য পদার্থের ক্ষয়কারী ক্ষতি সৃষ্টি করে। এতে তুলে পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসলফুরিক (HCl , HNO_3 , H_2SO_4), পাত্র ক্ষয় ($NaOH$, KOH), H_2O_2 , অক্সিজেন লিকার প্রভৃতি চূর্ণক ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কে, ক্ষয়কারী পদার্থের পিছনে রাখতে হবে।

উদাহরণের চূর্ণকটি হলো 10M H_2SO_4 চূর্ণক এবং এর থেকে 1M ঘনমাত্রার চূর্ণক প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে হবে।

1ম চলকের দ্রব্য হলো 100 ml অক্সিজেনমুক্ত চূর্ণক ও এর দ্রব্য পিঁপেট। সেমি-মাইক্রো অক্সিজেনমুক্ত চূর্ণকের আয়তন 100 ml। তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 চূর্ণক তৈরি করতে হবে।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

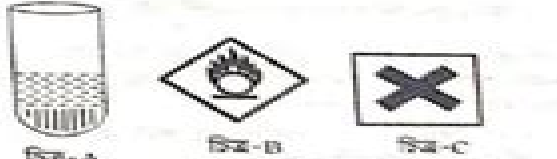
$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 চূর্ণকের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, পিঁপেটের সাহায্যে 10 ml চূর্ণক ঘনমাত্রার চূর্ণক নিয়ে তার সাথে অক্সিজেনমুক্ত চূর্ণকে 90 ml পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার চূর্ণক পাওয়া যাবে।

চিত্র-১১



১. অক্সিজেনমুক্ত কীভাবে বলা?
২. সাদা চূর্ণক K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যাঙ্গারকে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক চূর্ণকটির ব্যবহারকালে খান্সা কৃষ্টি হলে কবীর উপর সম্পর্কে বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

অক্সিজেনমুক্ত কী এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপ্ত যা মুদ্রণ বা অন্য উপাদানসমূহকে জারণ প্রতিরোধ করে থেকে রাখা করে রাখার সঞ্চেপ্ত করে।

২. সামান্য K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উত্তরটি ব্যাখ্যা করার জন্য নিম্নের উত্তরসূরী মাইক্রোপরি বিশ্লেষণ করি—

$$A + B \rightarrow C + D$$

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CH Cherryholmes

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