#### STD XII - CHEMISTRY

Time: 3 hrs.l.

MARCH - 2013

[Marks: 70

7

#### SECTION - I

Q. 1.	Select and write the most appropriate answers from the given	
	alternatives for each sub-question :	
	In hade control which the state of the state	

In body centred cubic structure the space occupied is about (b) 53% (d) 32%

(a) 68% (c) 38% (ii) For a gaseous reaction the unit of rate of reaction is

(a) L atm s<sup>-1</sup> (b) atm  $mol^{-1} s^{-1}$  $(c, stm s^{-1})$ 

(iii) Which of the following compounds contains S = O as well as S = S bonds?

(a) Sulphuric acid (b) Thiosulphuric aci (c) Sulphurous acid

(d) Thiosulphurous acid

(iv) Which of the following solutions shows maximum depression in freezing point? (a) 0.5 M Li<sub>2</sub>SO<sub>4</sub> (b) 1 M NaCl (c) 0.5 M Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> (d) 0.5 M BaCl<sub>2</sub>

(v) For a chemical reaction dS = -0.035 kJ/K and dH = -20 kJ. At what temperature does the reaction turn nonspontaneous? (a) 5.14 K (b) 57.14 K (d) 5714.0 K

(c) 571.4 K (vi) The standard e.m.f. of the following cell is 0.463 V

 $Cu_{(s)}$   $Cu_{(\alpha q)}^{++}$   $Ag_{(\alpha q)}^{++}$   $Ag_{(s)}$ . What is the standard potential of Cu electrode?

(a) 1.137 V (b) 0.337 V (c) 0.463 V

(d) -0.463 V

(vii) Fe<sub>2</sub>O<sub>3</sub> is reduced to spongy iron near the top of blast furnace by (a) H (b) CaO (c) SiO<sub>2</sub> (d) CO

### Q. 2. Answer any SIX of the following:

(i) Distinguish between crystalline solid and amorphous solid.

(ii) State Kohlrausch Law and write mathematical expression of molar conductivity of the given solution at infinite dilution.

(iii) Write cell reactions in lead storage battery during discharge.

(iv) Draw structures and write geometry of PCl<sub>3</sub> and PCl<sub>5</sub>.

(v) Prove that ΔH = ΔU + Δ nRT. What is the condition under which ΔU = ΔH?

(vi) Mention names and formulae of two ores of aluminium.

(vii) Derive the relationship between relative lowering of vapour pressure and molar mass of nonvolatile solute.

(viii) What is pseudo first order reaction? Give one example of it.

#### Q. 3. Answer any THREE of the following:

[9]

[12]

(i) Calculate the mole fraction and molality of HNO3 in a solution containing 12.2% HNO3. (Given - atomic masses: H = 1, N = 14, O = 16)

(ii) Consider the reaction —

 $3I_{(ag)}^{-} + S_2O_8^{2-} \longrightarrow I_{3(ag)}^{-} + 2SO_{4(ag)}^{2-}$  at particular time t,  $\frac{d[SO_4^{2-}]}{d} = 2.2 \times 10^{-2} \text{ M/s}.$ 

What are the values of the following at the same time?

(a) 
$$-\frac{d[I]}{dt}$$

(b) 
$$-\frac{d\left[S_2O_8^{2-}\right]}{dt}$$

(c) 
$$-\frac{d\left[I_3^-\right]}{dt}$$

# **Hsc 2013 Chemistry Question Old Syllabus**

**TD Snyder** 

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#### Table of Contents Hsc 2013 Chemistry Question Old Syllabus

- 1. Understanding the eBook Hsc 2013 Chemistry Question Old Syllabus
  - The Rise of Digital Reading Hsc 2013 Chemistry Question Old Syllabus
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc 2013 Chemistry Question Old Syllabus
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc 2013 Chemistry Question Old Syllabus
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc 2013 Chemistry Question Old Syllabus
  - Personalized Recommendations
  - Hsc 2013 Chemistry Question Old Syllabus User Reviews and Ratings
  - Hsc 2013 Chemistry Question Old Syllabus and Bestseller Lists

- 5. Accessing Hsc 2013 Chemistry Question Old Syllabus Free and Paid eBooks
  - Hsc 2013 Chemistry Question Old Syllabus Public Domain eBooks
  - Hsc 2013 Chemistry Question Old Syllabus eBook Subscription Services
  - Hsc 2013 Chemistry Question Old Syllabus Budget-Friendly Options
- 6. Navigating Hsc 2013 Chemistry Question Old Syllabus eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Hsc 2013 Chemistry Question Old Syllabus Compatibility with Devices
  - Hsc 2013 Chemistry Question Old Syllabus Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hsc 2013 Chemistry Question Old Syllabus
  - Highlighting and Note-Taking Hsc 2013 Chemistry Question Old Syllabus
  - o Interactive Elements Hsc 2013 Chemistry Question Old Syllabus
- 8. Staying Engaged with Hsc 2013 Chemistry Question Old Syllabus
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc 2013 Chemistry Question Old Syllabus
- 9. Balancing eBooks and Physical Books Hsc 2013 Chemistry Question Old Syllabus
  - $\circ$  Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc 2013 Chemistry Question Old Syllabus
- 10. Overcoming Reading Challenges
  - $\circ\,$  Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc 2013 Chemistry Question Old Syllabus
  - Setting Reading Goals Hsc 2013 Chemistry Question Old Syllabus
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc 2013 Chemistry Question Old Syllabus
  - Fact-Checking eBook Content of Hsc 2013 Chemistry Question Old Syllabus
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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