

For each compound, write
IONIC or **COVALENT**:

1

- a) MgCl_2
- b) CCl_4
- c) Na_2CO_3
- d) NH_3



Write the name of the
following **covalent** compounds:

2

- a) N_2O_3
- b) CBr_4
- c) P_2O_5



Write the number of
Valence electrons for each
element:

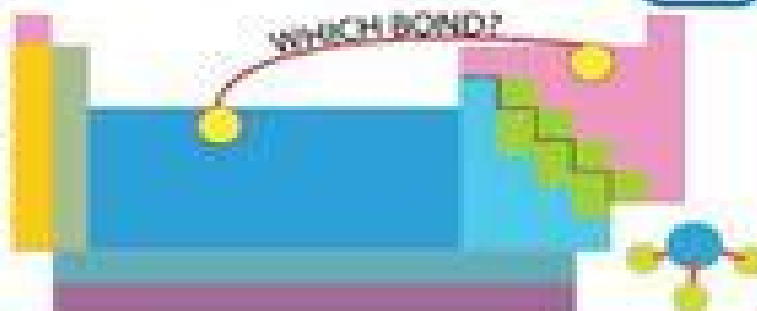
3

- a) **Phosphorus**
- b) **Helium**
- c) **Calcium**
- d) **Chlorine**



Ionic or Covalent?

4



Which Electron Dot Diagram
is **NOT** correct? **EXPLAIN**.

5



Write the **chemical formula**
of each compound:

6

- a) **Calcium oxide**
- b) **Iron (II) chloride**
- c) **Dihydrogen dioxide**



Holt Chemistry Bonds Compounds Concept Review Answers

Zhe Xu



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