problems (1) 1) w, = 26.60029 , w, = mass of empty crucible Wz = 26,7134 9 , wz = mass of Exicible and CaO War = W2 -W, = 26, 7134 - 26, 6002 = 0, 1132 9 Ca Ca O

H. wt-40,078 glowl H. wt-56.077 3/mol X = wt of Ca = $\frac{60.078 \times 0.1132}{36.077} = 0.08090$ 9 Strength - loo mL - strength of Ca = 0,08090 × 100 = 0,04045 g/100 ml sample 2) X= mass of Nacl , y = mass of Bacl , Again mass = 0,46379

Nacl X+ Y-0.2×369 , Again mass = 0,46379

Nacl X= 58,4491-1 H.wt.143,320/mil

w= X2

ut of Again X × 143.32 = 2,4524 X 9 Bacl ______ 2Ag Cl H.wt = 143, 3 2 g/mol x 2 M.wt - 208,23 glant m = 55 wit Je Agel 3x 143,32x2 =1,3766 y 9 = wt of Agel = 2,4524 X + 1,3766 y = 2.4524 X + 1.3766 (0,2356 - X) = 0,4637 9 - X = mass of Nacc = 0,12969 = Nacl %= 55,014. = y = mass of Back = 0,106 g . Baclzelo = 44,99.1.

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