

Use your notebook to complete all of the following review questions.

Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

- a) 75 g = 0.075 kg e) 4.2 kL = 420 000 cL
 b) 4530 cm = 0.0453 km f) 83.2 cm = 832 mm
 c) 0.8 A = 800 mA g) 975 mg = 0.975 g
 d) 3200 mA = 3.2 A h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	strongest	close together	least
liquid	weaker	farther apart	more
gas	weakest	farthest apart	most

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPOURATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change light given off
bubbles form heat produced
precipitate forms _____

Grade Nine Science Questions

Jan H. van Driel



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Table of Contents Grade Nine Science Questions

1. Understanding the eBook Grade Nine Science Questions
 - The Rise of Digital Reading Grade Nine Science Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Nine Science Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Nine Science Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Nine Science Questions
 - Personalized Recommendations
 - Grade Nine Science Questions User Reviews and Ratings
 - Grade Nine Science Questions and Bestseller Lists
5. Accessing Grade Nine Science Questions Free and Paid eBooks
 - Grade Nine Science Questions Public Domain eBooks
 - Grade Nine Science Questions eBook Subscription Services
 - Grade Nine Science Questions Budget-Friendly Options
6. Navigating Grade Nine Science Questions eBook Formats

- ePub, PDF, MOBI, and More
- Grade Nine Science Questions Compatibility with Devices
- Grade Nine Science Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Nine Science Questions
 - Highlighting and Note-Taking Grade Nine Science Questions
 - Interactive Elements Grade Nine Science Questions
- 8. Staying Engaged with Grade Nine Science Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade Nine Science Questions
- 9. Balancing eBooks and Physical Books Grade Nine Science Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Nine Science Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Nine Science Questions
 - Setting Reading Goals Grade Nine Science Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Nine Science Questions
 - Fact-Checking eBook Content of Grade Nine Science Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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