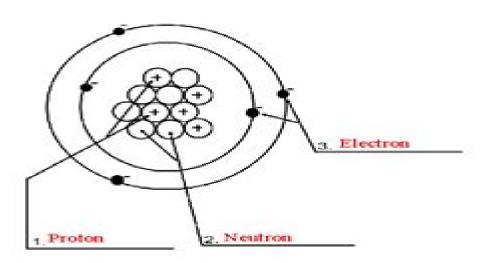
#### Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



- What type of charge does a proton have? Positive (+1)
- What type of charge does a neutron have? Neutral (0)
- What type of charge does an electron have? Negative (-1)
- 7. Which two subatomic particles are located in the nucleus of an atom?
  Protons and neutrons
- If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
- 9. What is the atomic number of the atom in the diagram above?
- 10. What is the atomic mass/mass number of the atom in the diagram above? (11+11=22) \*protons and neutrons added together) 22 is the atomic mass/mass number
- 11. How many protons are in the nucleus of an atom with an atomic number of 15?
  15 \*number of protons is the same as the atomic number
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
   Neutrons, atomic mass: 54.938 ~ 55 atomic number: 25,...55-25=30
- 14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
  Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
   Neutrons= 11 (atomic mass; 36 - 25 : atomic number)

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John S. Phillips, Cheryl Wistrom

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