

Ethnos und Demos in Tbilisi (Tiflis) - Armenier, Georgier und Russen in den Stadtdumawahlkämpfen 1890 -1897

Oliver Reisner

„Im Spannungsverhältnis zwischen dem neuzeitlichen Staatsbürgerbegriff als Definition politischer Gleichheit und der Verschiedenheit anderer sozialer und kultureller Solidaritätsgruppen einer Gesellschaft, von Ethnien, Religionen, Klassen usw. liegt der Kern der Minderheiten- und Nationalitätenproblematik“, beschreibt R. Lepsius den Widerspruch zwischen der Ebene des politisch verfaßten *Demos* und der des historischen, vielgestaltigen und (un-)interpretierbaren *Ethnos* einer Abstammungs- oder Kulturgemeinschaft. Zu Recht gibt er die Priorität dem verfassungsmäßig konkret definierten *Demos*.¹ Ebenso ist der Aussage zuzustimmen, daß ethnische Zugehörigkeit nicht immer das angemessenste Prinzip sei, nach welchem man soziale Aktivitäten oder Identitäten organisiert. Ethnizität kann nur unter bestimmten situativen Bedingungen als Ressource (v.a. als Deutungsmuster von Realität) einer sozialen, kulturellen oder rassistischen Gruppe zu ihrem Vorteil mobilisiert oder zu einem Hindernis werden.²

Wie komplex und problematisch sich das Verhältnis von *Ethnos* und *Demos* in einem extrem multinationalen Umfeld ohne kohärente, homogene Gemeinschaft mit gegenseitig abhängigen Interessen entfalten kann, soll im folgenden am Beispiel der Kaukasusmetropole Tiflis im späten Zarenreich aufgezeigt werden. Wir konzentrieren uns dabei auf die drei größten ethnischen Repräsentation in der Stadtduma. Denn gerade die Einführung demokratischer Prinzipien der städtischen Selbstverwaltung erforderte notwendigerweise die Definition dieses neuen Souveräns. Gleichzeitig werden mit demokratischen Institutionen erstmals auch Elemente des Wettbewerbs und der Unterstützung von Gruppen eingeführt, da der Zugang zu politischer Macht - wenn auch lokal und beschränkt - gerade vom Grad ihrer Mobilisierung abhängt. Wahlen sind deshalb auch nichts anderes als der Kampf um Unterstützung und Wettbewerb zwischen verschiedenen Gruppen. Ethnische bzw. nationale Definitionen bieten hierfür eine einfache und offensichtliche Basis, um Gruppenunterstützung her- bzw. sicherzustellen. Und damit wird auch - gewollt oder ungewollt - deren Politisierung ausgelöst.³

Auf den Rahmen des Zarenreiches kann hier nur bedingt eingegangen werden. Insofern die Konzentration der Kräfte im Zentrum und ihre Zersplitterung an der Peripherie typisch sind, können ethnische Grenzen „Sollbruchstellen“ dieser Zersplitterung werden, wenn sie mit sozialer Ungleichheit korrespondieren und von intellektuellen Eliten entsprechend als solche gedeutet werden.

¹ M.R. Lepsius, „Ethnos“ oder „Demos“. Zur Anwendung zweier Kategorien von Emmerich Francis auf das nationale Selbstverständnis der Bundesrepublik und auf die Europäische Einigung, in: ders., Interessen, Ideen und Institutionen, Opladen 1988, S. 247-255, hier S. 250.

² S. Wallman, Foreword and Introduction, in: ders. (Hg.), Ethnicity at Work, London/Basingstoke 1979, S. IX; zit. nach Fr. Heckmann, Ethnische Minderheiten, Volk und Nation. Soziologie innerethnischer Beziehungen, Stuttgart 1992, S. 32.

³ Siehe die Analyse des Verhältnisses von Demokratie und Nationalismus beim georgischen politischen Philosophen Ghia Nodia, Nationatism and Democracy, in: Journal of Democracy 3 (1992), H. 4, S. 3-22 und die Kommentare dazu von Francis Fukuyama (ebd., S. 23-28) und Shlomo Avineri (ebd., S. 28-31). Siehe dazu auch das Beispiel des Deutschen Reiches bei Stefan Invernitz, Wahlverhalten und Parteiensystem im Kaiserreich: einige quantitative Befunde, in: Heinrich Best (Hg.), Politik und Milieu. Wahl- und Elitenforschung im historischen und interkulturellen Vergleich, St. Katharinen 1989, S. 34-59, hier S. 35 (Hist.-Sozialwiss. Forschungen, Bd. 22).

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Table of Contents Ethnos Und Demos

1. Understanding the eBook Ethnos Und Demos
 - The Rise of Digital Reading Ethnos Und Demos
 - Advantages of eBooks Over Traditional Books
2. Identifying Ethnos Und Demos
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ethnos Und Demos
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ethnos Und Demos
 - Personalized Recommendations
 - Ethnos Und Demos User Reviews and Ratings
 - Ethnos Und Demos and Bestseller Lists
5. Accessing Ethnos Und Demos Free and Paid eBooks
 - Ethnos Und Demos Public Domain eBooks
 - Ethnos Und Demos eBook Subscription Services
 - Ethnos Und Demos Budget-Friendly Options

6. Navigating Ethnos Und Demos eBook Formats
 - ePub, PDF, MOBI, and More
 - Ethnos Und Demos Compatibility with Devices
 - Ethnos Und Demos Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ethnos Und Demos
 - Highlighting and Note-Taking Ethnos Und Demos
 - Interactive Elements Ethnos Und Demos
8. Staying Engaged with Ethnos Und Demos
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ethnos Und Demos
9. Balancing eBooks and Physical Books Ethnos Und Demos
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ethnos Und Demos
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ethnos Und Demos
 - Setting Reading Goals Ethnos Und Demos
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ethnos Und Demos
 - Fact-Checking eBook Content of Ethnos Und Demos
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was $\text{Ca(OH)}_2 \text{ s} + \text{Ca}^{2+} \text{ aq} + 2\text{OH}^- \text{ aq}$ the solubility product was determined to be $K_{\text{sp}} \text{ Ca(OH)}_2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{\text{sp}} \text{ Ca}_3\text{PO}_4 = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]$

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web lab report 8 determination of calcium hydroxide the purpose of this was to titrate a saturated Ca(OH)_2 solution with a standard HCl solution and find the Ksp or the OH concentration was found and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochloric acid

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest Ksp value to the theoretical Ksp for calcium hydroxide was the determination through pH method 2 introduction calcium hydroxide is a strong base soluble in water

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classified as insoluble based on solubility rules are actually slightly soluble each of these insoluble compounds actually dissolves to

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