#### Study Guide: Biochemistry exam 1

- Linetteen 2:
  - Define or describe the "Central Dagma" in relation to the storage & retrieval of genetic information, the creation of proteins and the expression of traits:
    - "Central Dogma" explains the storage, retrieval, & expression of genetic information.
    - Replication: duplication of DNA to make an identical copy.
    - Transcription: reading of a gene & its transcription into BNA.
    - Translation: translation of RNA sequence into the corresponding sequence of amino acids to form a protein.
    - "Information processing": Information flow from DNA to protein structure
      - DNA undergoes replication → transcribed into RNA → RNA translated into a protein → proteins develop into traits.
  - Illustrate and/or describe the general structural, physical and/or chemical characteristics of amino acids.
    - There are 29 standard amino acids.
    - Each amino acid has an amino group, and a certasyl group joined at the alpha cerbon.
    - The 'R' group is the unique feature for each amino acid.



- There's an O- on the corbosy's group by's the H+ goes to amino group, making NW+ at a certain pW.
- ◆ There are 4 main groups of omino acids:
  - Neutral non-polar; no chorpe, hydrophobic;
  - Neutral polars no charge, hydrophilic
  - Acidics extra carbonyl in R group, can be charged
  - Basic: extra amino in R group, can be + charged.

Charges occur in R groups:

- Amphoteric: when an amino acid can act as an acid or base (depending on the pH).
- Amino acids are "switterions" meaning they can carry both a + and charge on the same molecule.
- The alpha corbon is "asymmetric" if all 4 attached groups are different.
- Describe and/or illustrate the titration of an amino ocid, including the pi's or pKo's of ionizable groups.
  - Begin titration by adding a base, and watch the phi change.
  - You can increase the pill by adding a base, thus giving CDOII the time to give up its protos.
  - pilla reflects the dissociation of an acid.
  - pl is a point where 100% of the molecule has a 0 charge.

# **Essential Biochemistry Study Guide**

Marcia L. Gillette,Ira Blei,Wendy Gloffke,George Odian  ${\bf Essential\,\,Biochemistry\,\,Study\,\,Guide:}$ 

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# **Table of Contents Essential Biochemistry Study Guide**

- 1. Understanding the eBook Essential Biochemistry Study Guide
  - The Rise of Digital Reading Essential Biochemistry Study Guide
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Essential Biochemistry Study Guide
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Essential Biochemistry Study Guide
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Essential Biochemistry Study Guide
  - Personalized Recommendations
  - Essential Biochemistry Study Guide User Reviews and Ratings
  - Essential Biochemistry Study Guide and Bestseller Lists

- 5. Accessing Essential Biochemistry Study Guide Free and Paid eBooks
  - Essential Biochemistry Study Guide Public Domain eBooks
  - Essential Biochemistry Study Guide eBook Subscription Services
  - Essential Biochemistry Study Guide Budget-Friendly Options
- 6. Navigating Essential Biochemistry Study Guide eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Essential Biochemistry Study Guide Compatibility with Devices
  - Essential Biochemistry Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Essential Biochemistry Study Guide
  - Highlighting and Note-Taking Essential Biochemistry Study Guide
  - Interactive Elements Essential Biochemistry Study Guide
- 8. Staying Engaged with Essential Biochemistry Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Essential Biochemistry Study Guide
- 9. Balancing eBooks and Physical Books Essential Biochemistry Study Guide
  - Benefits of a Digital Library
  - $\circ$  Creating a Diverse Reading Collection Essential Biochemistry Study Guide
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Essential Biochemistry Study Guide
  - Setting Reading Goals Essential Biochemistry Study Guide
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Essential Biochemistry Study Guide
  - Fact-Checking eBook Content of Essential Biochemistry Study Guide
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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