

TOPIC 5 FURTHER MECHANICS

5A Further momentum

Question number	Answer	Additional guidance	Mark
1	B	Note that while D can be true in some circumstances, it is not true if the vehicles' speeds were different initially.	(1)

(Total for Question 1 = 1 mark)

Question number	Answer	Additional guidance	Mark
2(a)	$250 \times 2.4 = 600$ (1) kg m s^{-1} (1)	Do not allow kg m/s.	(2)
2(b)	Momentum decrease (1) Change in momentum of probe = (-)change in momentum of gas (in opposite direction) (1) Gas momentum change is 'forwards' so probe change is 'backwards'. (1)	An answer in terms of Newton's third law should be credited, provided students relate it to momentum and velocity. Allow sum of momentum <u>change</u> = 0 Do not allow <u>total</u> momentum = 0	(3)
2(c)	$Ft = \Delta p$ $F = \frac{600}{20}$ (1) $F = 30 \text{ N}$ (1)	Allow error carried forward from 2(a).	(2)
2(d)	Average mass of probe will be lower. (1) As momentum change is the same, $\Delta v = \frac{\Delta p}{m}$ will be greater. (1)		(2)
2(e)(i)	Momentum change of probe would double. (1) Momentum change of gas = Δmv (for the same mass): double v , Δp will double. (1)		(2)

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this

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october 8 2014 summary the purpose of this experiment is to be able to understand the relationship between the pressure
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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be
22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k
for temperature into the ideal gas equation and solve for r r

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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