

Chemistry: NCERT Multiple Choice Questions

Multiple Choice Questions

Structure the solution of each question by using the appropriate and appropriate equation.

Figure 1: A graph showing the relationship between the concentration of a solution and its boiling point.

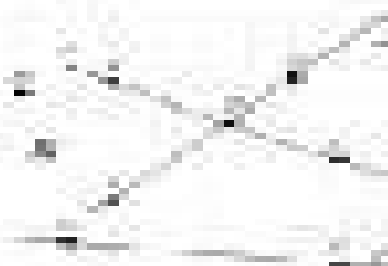


Figure 2: A graph showing the relationship between the concentration of a solution and its boiling point.



1. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

2. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

3. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

4. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

5. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

6. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

7. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

8. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

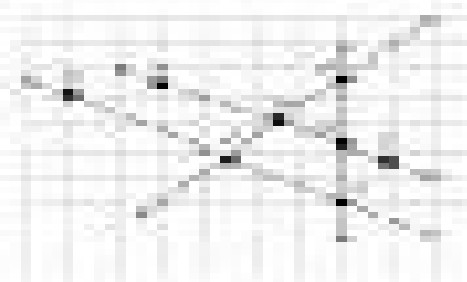
- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

9. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

10. The boiling point of a solution is 100°C. The concentration of the solution is 10 g/100g. The boiling point of pure water is 100°C. The boiling point of the solution is 100°C.

- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C



- a. 100°C
- b. 100°C
- c. 100°C
- d. 100°C

Geometry Eoc Winter Packet Answers

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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