

## Polyatomic Ions Worksheet

### Polyatomic Ions

I. Name or write the formula for the following polyatomic ions:

sulfate	_____	$\text{CO}_3^{2-}$	_____
nitrite	_____	$\text{SO}_4^{2-}$	_____
perchlorate	_____	$\text{BrO}_3^-$	_____
hypochlorite	_____	$\text{CO}_3^{2-}$	_____
chlorate	_____	$\text{PO}_4^{3-}$	_____
phosphate	_____	$\text{ClO}^-$	_____
percarbonate	_____	$\text{NO}_2^-$	_____
chromate	_____	$\text{PO}_4^{3-}$	_____
hyposulfite	_____	$\text{FeO}_4^{2-}$	_____
permanganate	_____	$\text{SO}_4^{2-}$	_____
carbonate	_____		

II. Name or write the formula for the following Type I polyatomic ionic compounds:

barium hydrosulfide	_____	$\text{Ba}(\text{OH})_2$	_____
sodium nitrite	_____	$\text{Ca}(\text{OH})_2$	_____
ammonium chloride	_____	$\text{Ag}_2\text{SO}_4$	_____
calcium bisulfate	_____	$\text{MgCO}_3$	_____
rubidium perchlorate	_____	$\text{Sr}_2\text{FeO}_4$	_____
chromium sulfate	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	_____
aluminum acetate	_____	$\text{SrSO}_4$	_____
ammonium nitrate	_____	$\text{Fe}(\text{NO}_3)_3$	_____
magnesium hydrogen carbonate	_____	$\text{Ca}(\text{HCO}_3)_2$	_____
silver hypochlorite	_____	$\text{Sr}_2\text{CO}_3$	_____
potassium cyanide	_____	$\text{NaNO}_2$	_____
barium iodate	_____	$\text{Ba}(\text{OH})_2$	_____

# Lots Of Ionic Naming Practice Problems Key

**Chris Hren, Peter J. Mikulecky**



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