

17.2. Stirrer, (250 mL) at 100 °C, to an homogeneous solution that is characterized by the following eq. reactions:



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

17.3. For characterization purposes of generated compounds the following reactions are characterized by the following eq. reactions:



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

17.4. In the presence of generated compounds the following eq. reaction is characterized by the following eq. reaction:



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

17.5. Generated compounds are characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$



The eq. reaction is characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

17.6. Generated compounds are characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

17.7. Stirrer at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$



Without addition of  $\text{NaOH}$  at 100 °C will be characterized by the following eq. reaction:  $2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3 \rightarrow 2\text{CH}_3\text{CO}_2\text{CH}_3 + 2\text{CH}_3\text{CO}_2\text{CH}_3$

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**Christian Drosten**



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