

Total No. of Questions : 8]

P303

SEAT No. :

[Total No. of Pages : 2

**[4763]Ext. - 11**  
**F.Y.B.Com (External)**  
**MANAGERIAL ECONOMICS**  
**(2013 Pattern) (Optional)**

*[Time : 3 Hours]*

*[Max. Marks : 100*

*Instructions to the candidates:*

- 1) Question number one is compulsory.*
- 2) Attempt Five Questions from remaining.*
- 3) Figures to the right indicate full marks.*

- Q1)* State and explain scope and characteristics of managerial Economics. **[20]**
- Q2)* Explain determinants of demand **[16]**
- Q3)* Explain methods of Business forecasting. **[16]**
- Q4)* State and explain the law of variable properties. **[16]**
- Q5)* Explain usefulness of Capital Budgeting. **[16]**
- Q6)* Explain use and limitations of Cost Benefit analysis. **[16]**
- Q7)* State and explain role of MNCs. **[16]**
- Q8)* Write notes. **[16]**
- a) Business trends in India.
  - b) Objectives of the firm

*P.T.O.*

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Fybcom Accounts Pune University Paper Pattern 2013**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web goals understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1

$\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{O l}$   $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{O l}$  one mole of hcl would be fully neutralized by one mole of naoh

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide naoh and oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak  $\text{pK}_a$  of an unknown acid or  $\text{pK}_b$  of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance  $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$  to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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ascertain the concentration of the naoh

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction

naoh aq c6h4 cooh cook aq c6h4 coona cook aq h2o l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the