

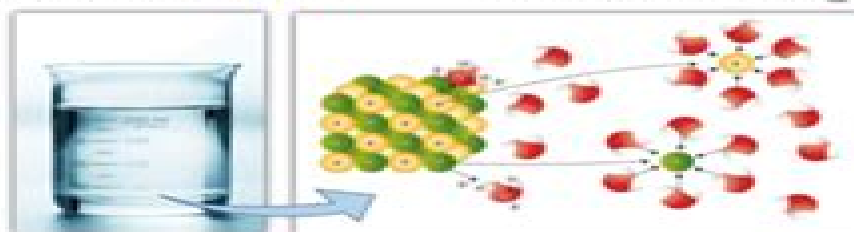
Chem102 Preparation of Solutions

HANDOUT 1

Solutions are homogenous mixtures that contains two or more substances. As we all know, solutions are homogenous combination of two or more compounds. When a solution is homogenous, the difference between the physical property of both solutions is cannot be distinguished. A solution has uniform composition. Examples are sugar and table salt in water. The dissolving medium (water) is the solvent and the dissolved substances (sugar or salt) is the solute.

SOLUTE **NaCl**

SOLVENT **H₂O**



Components:

1. Solvent
 - present in the greatest quantity
 - determines the phase of solution
 - if water is a component, it is usually the solvent.
2. Solute
 - component dissolved in the solvent

If the SOLUTE SOLVENT interaction is stronger than the solute-solute interaction and solvent solvent interaction then the solution process would be spontaneous. Water is a universal solvent but not all substances can dissolved by water.

Solubility is a temperature dependent

- maximum amount of solute that dissolves in a given amount of solution or solvent at a given temperature
- unit = grams of solute per mL of solvent

E.g. Solubility of NaCl = 35.7 g/100 mL at 20 °C

Futa Chem102 Material

Terry C. Jones



Futa Chem102 Material:

This book delves into Futa Chem102 Material. Futa Chem102 Material is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Futa Chem102 Material, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Futa Chem102 Material
- Chapter 2: Essential Elements of Futa Chem102 Material
- Chapter 3: Futa Chem102 Material in Everyday Life
- Chapter 4: Futa Chem102 Material in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Futa Chem102 Material. The first chapter will explore what Futa Chem102 Material is, why Futa Chem102 Material is vital, and how to effectively learn about Futa Chem102 Material.
3. In chapter 2, this book will delve into the foundational concepts of Futa Chem102 Material. This chapter will elucidate the essential principles that must be understood to grasp Futa Chem102 Material in its entirety.
4. In chapter 3, this book will examine the practical applications of Futa Chem102 Material in daily life. This chapter will showcase real-world examples of how Futa Chem102 Material can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Futa Chem102 Material in specific contexts. This chapter will explore how Futa Chem102 Material is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Futa Chem102 Material. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Futa Chem102 Material.

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