



English Language A Level

Examination and Revision Information

Examination Board:	AQA A Level English Language
Number of Examination Papers:	<u>Paper 1: Language, the Individual and Society</u> What's assessed: - Textual variations and representations - Children's language development (0-11 years) - Methods of language analysis are integrated into the activities <u>Paper 2: Language Diversity and Change</u> What's assessed: - Language diversity and change - Language discourses - Writing skills - Methods of language analysis are integrated into the activities
Structure of Examination Papers:	<u>Paper 1</u>: written exam: 2 hours 30 minutes. 100 marks. 40% of A-level <u>Paper 2</u>: written exam: 2 hours 30 minutes. 100 marks. 40% of A-level
Coursework:	<u>Non-exam assessment: Language in Action</u> What's assessed: - Language Investigation - Original Writing - Word count: 3,500. 100 marks. 20% of A-level. Assessed by teachers. Moderated by AQA
Key Information:	- The English Language A Level course is a two year, linear curriculum. - The qualification is 80% examination based and 20% coursework. - Five Assessment Objectives cover each area of the course.
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What are the best ways to revise for this subject?	- Self-Testing: Use flash cards to cover each key area for revision. It is a good idea to create terminology grids to revise the frameworks and theory revision maps to cover each area of language diversity and language change. - Your file notes: revise by using all of the notes you have taken in class. These are your best revision resource. - Teacher feedback: review all of your assessed work and look at the feedback your teachers have given for improvement. Use this as a starting point in order to write your next essay. - Past Papers: Answer individual questions from past A Level papers. Make sure you know the question types and timings. - Quizlet is a useful resource for checking your knowledge of terminology and theorists.



English Language As Level Revision Categorizing Texts

Annelies Wilder-Smith



English Language As Level Revision Categorizing Texts:

Unveiling the Power of Verbal Art: An Emotional Sojourn through **English Language As Level Revision Categorizing Texts**

In a global inundated with monitors and the cacophony of fast conversation, the profound power and emotional resonance of verbal art usually disappear in to obscurity, eclipsed by the continuous onslaught of noise and distractions. However, situated within the musical pages of **English Language As Level Revision Categorizing Texts**, a interesting perform of fictional beauty that impulses with fresh thoughts, lies an remarkable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, this enchanting opus courses visitors on a psychological odyssey, delicately revealing the latent potential and profound impact embedded within the complex internet of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is key themes, dissect its fascinating writing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655
 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1
 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide (NaOH) and oxalic acid ($H_2C_2O_4 \cdot 2H_2O$) and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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the required shape of metal is obtained by removing the unwanted material from the work piece in the form of chips is called

cutting shaping example turning drilling milling boring etc 2 classify the relative motion between work piece and cutting tool
i