

5. Complete the table below as shown by the example. Combine the ions into a correct formula, and name the compound.

	$\text{CN}^-$	$\text{Cr}_2\text{O}_7^{2-}$	$\text{P}^{3-}$
$\text{K}^+$	$\text{KCN}$ potassium cyanide		
$\text{Zn}^{2+}$			
$\text{Cr}^{3+}$			

6. (optional) Complete the table below as shown by the example. Combine the ions into a correct formula, and name the compound.

	chlorate ion	sulfate ion	phosphate ion
cuprous ion	$\text{CuClO}_3$ cuprous chlorate		
nickel(II) ion			
mercurous ion			
lead(IV) ion			
barium ion			

# Laboratory Manual Prentice Hall Chemistry Answers

**SA Adler**



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