

Report FOR EXPERIMENT # 1: DETERMINATION OF MELTING POINTS

Your name

TA's name

Your Partner's name

Lab Section

RESULTS:

Melting point of urea: 134-154 °C

Melting point of trans-cinnamic acid: 133-134 °C

Melting range of a mixture of urea and trans-cinnamic acid: 110-125 °C

Melting point of unknown: 133-133.5 °C

Melting point of unknown mixed with urea: 112-123 °C

Melting point of unknown mixed with Melting point of unknown: 133-133.5 °C

Identity of unknown: Melting point of unknown: 133-133.5 °C

CONCLUSION:

In this lab we used melting points to determine the purity of organic compounds. Pure compounds were seen to have a narrow melting range, while impurities were seen to cause the melting range to become lower and broader. From the mixture melting points we were definitively able to identify our unknown as trans-cinnamic acid.

QUESTIONS: DETERMINATION OF MELTING POINTS

1. What is a safety hazard associated with this lab session?

One will be using instruments that will be heated to high temperatures. So, one should take precautions when handling these instruments.

Lab Report On Melting Points

Daniel R. Palleros



Lab Report On Melting Points:

Report Kansas State Board of Health,1912 **Practical Lab Manual of Pharmaceutical Organic Chemistry - I** Dr. Priyanka Gupta Manglik,2024-08-15 Includes a collection of foundational organic chemistry experiments for pharmaceutical studies with step by step procedures observations and safety guidelines Experimental Organic Chemistry Daniel R. Palleros,2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion

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and the use of technology in science learning and teaching K 16 The volume investigates issues and concerns related to this theme through various perspectives addressing design research professional practice and evaluation Beginning with definitions the historical evolution and policy guiding these learning experiences are explored from several viewpoints Effective design and implementation of laboratory work and technology experiences is examined for elementary and high school classrooms as well as for undergraduate science laboratories informal settings and science education courses and programs In general recent research provides evidence that students do benefit from inquirybased laboratory and technology experiences that are integrated with classroom science curricula The impact and status of laboratory and technology experiences is addressed by exploring specific strategies in a variety of scientific fields and courses The chapters outline and describe in detail researchbased best practices for a variety of settings **Bulletin of the Hygienic Laboratory** ,1912

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Laboratory Manual for Principles of General Chemistry Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Bulletin Kansas State Board of Health,1911 *The Journal of Industrial and Engineering Chemistry* ,1913 Energy Research Abstracts ,1981-10 **Journal of Industrial and Engineering Chemistry** ,1922 **Nuclear Safety** ,1965 **The Student's Lab Companion** John W. Lehman,2008 This comprehensive lab companion provides enough theory to help students understand how and why an operation works but emphasizes the practical aspects of an operation to help them perform the operation successfully in the lab For undergraduate or graduate students taking organic chemistry lab This comprehensive lab companion provides enough theory to help students understand how and why an operation works but emphasizes the practical aspects of an operation to help them perform the operation successfully in the lab The Second Edition makes substantive revisions of many operations to clarify existing material and add new information More environmentally friendly i e green lab experiments are encouraged Ideal for professors who write their own lab

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