

Ionic and Covalent Naming Practice **Answers**

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

- | | | |
|-----|---------------------------|----------------------------------|
| 1) | Na_2CO_3 | sodium carbonate |
| 2) | P_4O_{10} | tetraphosphorus decaoxide |
| 3) | N_2O_5 | dinitrogen pentoxide |
| 4) | FeSO_4 | iron (II) sulfate |
| 5) | SiO_2 | silicon dioxide |
| 6) | FeCl_3 | iron (III) chloride |
| 7) | CoBr_2 | cobalt (II) bromide |
| 8) | B_2H_4 | diboron tetrahydride |
| 9) | CO | carbon monoxide |
| 10) | PCl_3 | phosphorus trichloride |

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

- | | | |
|-----|------------------------|--|
| 11) | dinitrogen trioxide | N_2O_3 |
| 12) | nitrogen dioxide | NO_2 |
| 13) | sulfur pentoxide | SO_5 |
| 14) | lithium acetate | $\text{LiC}_2\text{H}_3\text{O}_2$ |
| 15) | phosphorus trifluoride | PF_3 |
| 16) | vanadium (V) oxide | V_2O_5 |
| 17) | aluminum hydroxide | $\text{Al}(\text{OH})_3$ |
| 18) | zinc sulfide | ZnS |
| 19) | silicon tetrafluoride | SiF_4 |
| 20) | silver phosphate | Ag_3PO_4 |

Guided Answers Ionic Bonds Ionic Compounds

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