

Name _____
Chemistry I-2 HONORS Study Guide for Acids and Bases (Chapter 17)

Students should be able to...

- Know properties of acids and bases
- Know definitions of acids and bases according to Arrhenius, Bronsted-Lowry, and Lewis.
- Label Bronsted-Lowry acid-base reactions as acid, base, conjugate acid, and conjugate base
- Identify polyprotic acids and bases.
- Review naming of acids, both binary and ternary.
- Calculate pH, pOH, H⁺ concentration, and OH⁻ concentration for either a strong or weak acid or base.
- Calculate the solute and solvent necessary for a dilution
- Calculate the volume necessary to neutralize an acid/base
- Measure the pH and acidity/alkalinity of household chemicals (laboratory)
- Calculate the % by mass of acetic acid in vinegar (laboratory)

Review problems

1. List three properties of acids.
2. List three properties of bases.

Fill in the table concerning definitions of acids and bases.

Definition	Acid	Base
Traditional (Arrhenius)		
	Electron pair acceptor	
		Proton acceptor

Name the following acids. Circle any polyprotic acids.

9. H₃PO₄ _____
10. HF _____
11. H₂S _____
12. HNO₃ _____
13. HClO₄ _____

Solve.

14. [H⁺] = 5.32×10^{-4}
[OH⁻] = _____
pH = _____
pOH = _____
Acid or Base? _____

Guided And Study Workbook Acids And Bases

Michael Seilmaier



Guided And Study Workbook Acids And Bases:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises Critical Care Study Guide Gerard J. Criner, Gilbert E. D'Alonzo, 2013-03-09 Critical care medicine is a dynamic and exciting arena where complex pathophysiologic states requiring extensive knowledge and clinical acumen are commonly found Caring for critically ill patients requires an extensive knowledge of basic pathophysiology as well as awareness of the appropriate diagnostic tests and therapeutic interventions Because this knowledge base crosses many different disciplines introduction to caring for the intensive care patient while exciting may also be intimidating This textbook is designed toward making the dynamic environment of the critical care unit understandable and the approach to the patient both logical and successful The book contains three components 1 description of the procedural tasks commonly performed for the critically ill patient 2 explanation of the most common pathophysiologic states encountered and 3 description of specific disease entities with details of their differential diagnosis diagnostic strategy and therapeutic plan This book also uses several educational approaches that we have found useful in our own teaching sessions and prior textbooks Clinical cases introduce chapters and highlight chapter segments to emphasize clinical relevancy As an additional study aid margin notes highlight important teaching points and facilitate easy review of chapter content To consolidate the principles outlined in each chapter review questions with full text explanations are provided at the end of each chapter All these elements help reinforce the most important messages for the reader Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The

authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us **Organic**

Chemistry, 13e Student Study Guide and Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A.

Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text *Qualitative Chemical Analysis. A Guide in the Practical Study of Chemistry and in the Work of Analysis ... Second Edition, Revised* Silas H. DOUGLAS (and PRESCOTT (Albert B.)), 1876 **Barron's Science 360: A**

Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A.

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