Common Constants		
Avogadro's mumber (NG)	6.022 x 10 ²³ / mol	
Charge on an electron (s)	1.60 x 10 ° C	
Faraday's Constant (F)	9.648 X 10 ⁴ C / mol	
Gas Constant (X)	0.08206 (1. ostol) 8.31 / (mart K)	
Mass of an electron	5.486 X10 ⁻⁴ amsu	
Mass of a neutron	1.00 amu 1.67 X 10 ⁻²⁴ g	
Mass of a proton	1.00 amu 1.67 X 10 ⁻²⁴ g	
Pi	3.1415927	
Planck's Courtant (iii)	6,626 X 10 ⁻¹⁴ J/s	
peed of Light in a Vacuum (r)	2.99 X 10° m/s	

Common Con	version Factors	
Units of Length		
meter (m)	1 m = 39.37 in = 1.094 yards	
inch (in)	1 m = 2.54 cm	
angstrom (A)	1 A = 10 ⁻⁸ cm = 10 ⁻¹⁰ m	
Units of Volume	U.S.	
liter (L)	1 L = 0.001 m ³ = 1000 cm ³	
millimeter (mL)	1 mL = 0.001 L = 1 cm ³	
Units of Mass		
gram (g)	1 g = 0.001 g	
pound (Ib)	1 0b = 0.4535924 kg	
atomic mass unit (amu)	1 amu = 1.6605389 x 10 ⁻⁵⁴ kg	
Units of Energy		
4.184 joule (F) = 1 ther	rmochemical calorie (cal)	
Liter atmosphere =	24,217 cat = 101,325 J	
Nutritional calorie (C	Cal) = 1000 cal = 4184 J	
Units of Pressure		
1 tocr =	1 mmHg	
1 puscal (Pa) = 1	$N/m^2 = 1 \text{ kg/m} \cdot s^2$	
1 atmosphere (atm) = 760 mmHg =	760 toer = 101,325 N·m² = 101,325 Pr	
1 bar = 10° Pr	a = 10 ⁵ km/m •s ⁵	

SI Units				
Unit Name	Unit Symbol	Quantity Name	Quantity	
meter	200	length	1, 8, 7	
kilogram	kg	disable in	ma.	
second		Sagare	100	
воореее	A	electric current	1	
Kelvin	к	thermodynamic temperature	T	
candela	ed	Suminous intensity	I.	
mole	mol	amount of substance	10.	

Prefixes - Metric Units				
Proofix	Symbol	Magnitude		
ENEX-	G	1×10^{9}		
miega-	3.4	3 x: 104		
kile-	*	1×10^{2}		
	Base Unit	- 10		
centi-		1×10^{-4}		
milli-		1×10^{-3}		
majoro-	P.	1×10^{-4}		
name-		1×10^{-9}		

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Metad	Ion Found
Lithium	Lan
Potassium	100°
Cadcinon	Ca ³
S-e-dissens	Na*
Magnesium	Machine
A homimom	All ^a
Zanc	Zn2~
Chromium	Crite
Iron	Fe3-
Nacioeli	2697
Tim	Sp.2**
Lead	P45-7**
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