

**KEEP IT SIMPLE SCIENCE***Chemistry Module 5*

Chemical Equilibrium

WORKSHEETS

Worksheet 1**The Basics**

Student Name.....

1.
a) When any chemical reaction reaches an equilibrium in a closed system, what becomes true about the concentration of all the chemical species present?

b) Explain why this is so, in the case of a:
i) static equilibrium.

ii) dynamic equilibrium.

2.
What are the thermodynamic reasons why some reactions (eg combustion) are never reversible?

3.
What is meant by "open" or "closed" chemical systems?

4.
Why can a dynamic equilibrium NOT occur in an open system?

5.
Write an ionic equation to represent the dynamic equilibrium reached between the solid and the dissolved ions in each of these species, when a solution reaches saturation.

a) potassium bromide

b) magnesium chloride

c) iron (III) nitrate

d) ammonium sulfate

e) calcium carbonate

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LP Steffe

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the 'LP Steffe' text.

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Basic Thermochemistry in Materials Processing Gabriel Plascencia, David Jaramillo, 2017-02-28 This book provides the reader with some thermochemistry notes The intention is to provide a simple easy to understand text which serves as a complimentary material to more complex books It also provides students and those beginning in the field with several application examples used in different areas of materials processing The book presents fully solved problems some quite often found in major metallurgical operations

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Human-computer Interaction, INTERACT '99 Martina Angela Sasse, Chris W. Johnson, 1999 This text provides an overview of leading edge developments in the field of human computer interaction It includes contributions from many key areas that are influencing the use of computers Sections include speech technology interaction with mobile and hand held computers e business web based systems virtual reality and haptic interfaces **Resources in Education**, 1982 **Excel HSC Chemistry** Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date coverage of the core topics plus 3 Option topics Industrial Chemistry Shipwrecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to section the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second hand and Investigations and Problem Solving %this way you will be able to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for

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