



Google Play

# Google Play Manual

**Didier Musso**



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web the above equation works only for neutralizations in which there is a 1 1 ratio between the acid and the base the example below demonstrates the technique to solve a titration problem for a titration of sulfuric acid with sodium hydroxide

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web titrant and analyte is a pair of acid and base acid base titrations are monitored by the change of pH as titration progresses let us be clear about some terminologies before we get into the discussion of titration curves titrant solution of a known concentration which is added to another solution whose concentration has to be determined

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