

## CHEMISTRY PAPER-II

## OBJECTIVE

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

**Note:** You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?  
 (A) *Be* (B) *Ra* (C) *Ba* ☒ (D) *Rn*
- (2) Which element forms an ion with charge +3?  
 (A) *Be* ☒ (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:  
 (A) *N<sub>2</sub>O* (B) *N<sub>2</sub>O<sub>3</sub>* ☒ (C) *N<sub>2</sub>O<sub>4</sub>* (D) *N<sub>2</sub>O<sub>5</sub>*
- (4) Chlorine heptaoxide (*Cl<sub>2</sub>O<sub>7</sub>*) reacts with water to form:  
 (A) Hypochlorous acid (B) Chloric acid ☒ (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO<sub>4</sub>* is:  
 (A) *ClO<sub>3</sub>* (B) *ClO<sub>2</sub>* (C) *Cl<sub>2</sub>O<sub>3</sub>* ☒ (D) *Cl<sub>2</sub>O<sub>7</sub>*
- (6) Which of the following is a non-typical transition element?  
 (A) *Cr* (B) *Mn* ☒ (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:  
 (A) Position isomerism (B) Functional group isomerism ☒ (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:  
 (A) Halogenation ☒ (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?  
☒ (A) *AlCl<sub>3</sub>* (B) *HNO<sub>3</sub>* (C) *BeCl<sub>2</sub>* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?  
 (A) *E<sub>1</sub>* and *E<sub>2</sub>* (B) *E<sub>2</sub>* and *S<sub>N</sub>2* (C) *S<sub>N</sub>1* and *E<sub>2</sub>* ☒ (D) *E<sub>1</sub>* and *S<sub>N</sub>1*
- (11) Which compound is more soluble in water?  
☒ (A) *C<sub>2</sub>H<sub>5</sub>OH* (B) *C<sub>6</sub>H<sub>5</sub>OH* (C) *CH<sub>3</sub>COCH<sub>3</sub>* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?  
 (A) Methanal (B) Ethanal (C) Propanal ☒ (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?  
☒ (A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?  
 (A)  $\text{H}_2/\text{Ni}$  (B)  $\text{H}_2/\text{Pt}$  (C) *NaBH<sub>4</sub>* ☒ (D) *LiAlH<sub>4</sub>*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:  
 (A) Alcoholic *KOH* ☒ (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO<sub>3</sub>*
- (16) Most concentrated solid nitrogen fertilizer is:  
 (A) *NH<sub>3</sub>* (B) *(NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>* ☒ (C) *(NH<sub>2</sub>)<sub>2</sub>CO* (D) *NH<sub>4</sub>NO<sub>3</sub>*
- (17) Mark the correct statement.  
☒ (A) *Na<sup>+</sup>* is smaller than *Na* atom (B) *Na<sup>+</sup>* is larger than *Na* atom  
 (C) *Cl<sup>-</sup>* is smaller than *Cl* atom (D) *Cl<sup>-</sup>* ion and *Cl* atom are equal in size

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