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Chapter 3,

Finding Percent by Mass

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Hydrogen	1.30 g hydrogen	$\frac{1.30 \text{ g hydrogen}}{20.00 \text{ g sucrose}} \times 100\% =$   6.50% hydrogen
Oxygen	10.26 g oxygen	$\frac{10.26 \text{ g oxygen}}{20.00 \text{ g sucrose}} \times 100\% =$   51.30% oxygen
Total	20.00 g sucrose	 42.2% 51.30% 6.50%

Math Skills Transparency Answers 15

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