

PART - I

15 X 1 = 15

- I Answer all the questions.
- Which of the following is / are true with respect to Carbon - 12  
a) relative atomic mass is 12u b) oxidation number of Carbon is +4 in all its compounds c) mole of Carbon - 12 contain  $6.022 \times 10^{22}$  carbon atoms d) all of these
  - How many electrons in an atom with atomic number 105 can have  $(n + l) = 8$   
a) 30 b) 17 c) 15 d) unpredictable
  - Among the following which is isoelectronic with  $O^{2-}$  ion,  
a)  $Cl^-$  b)  $F^-$  c)  $Ca^{2+}$  d)  $N^{2-}$
  - The cause of permanent hardness of water is due to  
a)  $Ca(HCO_3)_2$  b)  $Mg(HCO_3)_2$  c)  $CaCl_2$  d)  $MgCO_3$
  - Sodium is stored in  
a) Alcohol b) Water c) Kerosene d) None of these.
  - It temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes  
a) 4p b) 2p c) p d) 3p
  - Heat of combustion is always  
a) Positive b) negative c) zero d) either positive or negative
  - In the reaction,  $Fe(OH)_3(s) \rightleftharpoons Fe^{3+}(aq) + 3OH^-(aq)$  if the concentration of  $OH^-$  ions is decreased by  $1/4$  times, then the equilibrium concentration of  $Fe^{3+}$  will  
a) not changed b) also decreased by  $1/4$  times  
c) increase by 4 times d) increase by 64 times
  - According to Raoult's law, the relative lowering of vapour pressure for a solution is equal to  
a) mole fraction of solvent b) mole fraction of solute  
c) number of moles of solute d) number of moles of solvent
  - Number of sigma ( $\sigma$ ) and pi ( $\pi$ ) bonds in acetylene is .....  
a)  $2\sigma$  &  $2\pi$  b)  $3\sigma$  &  $3\pi$  c)  $2\sigma$  &  $3\pi$  d)  $3\sigma$  &  $2\pi$
  - Sodium nitropruside reacts with sulphide ion to give a purple colour due to the formation of  
a)  $[Fe(CN)_5 NO]^{3-}$  b)  $[Fe(NO)_5 CN]^+$  c)  $[Fe(CN)_5 NOS]^{4-}$  d)  $[Fe(CN)_5 NOS]^{3-}$
  - What is the hybridisation state of benzyl carbonium ion?  
a)  $sp^2$  b)  $sp^2d$  c)  $sp^3$  d)  $sp^2d$
  - ..... is used for welding and cutting metals.  
a) Ethylene b) Propylene c) Poly ethylene d) Oxy - acetylene
  - What should be the correct IUPAC name of diethyl chloromethane?  
a) 3 - chloro pentane b) 1 - chloropentane  
c) 1 - chloro - 1, 1, diethyl methane d) 1 - chloro - 1 - ethylepropane
  - The pH of normal rain water is .....  
a) 6.5 b) 7.5 c) 5.6 d) 4.6

PART - II

6 x 2 = 12

- II Answer any six. (q.No. 24 is compulsory).
- Define modern periodic law.
  - What is water - gas shift reaction?

# Grade 11 Chemistry Half Yearly Exam

**J Dewey**



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