

## GRADE 11 PHYSICAL SCIENCES MARCH TEST

### QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER (1.1-1.6) on your folio paper. Example 1.7: D.

- 1.1 The graphs below show the relationship between the net force and the acceleration for two masses X and Y.



Which one of the following statements is true?

- A. Mass of X is greater than mass of Y.  
B. Acceleration of X is greater than acceleration of Y.  
C. Mass of Y is greater than mass of X.  
D. The slope of X is greater than the slope of Y.
- 1.2 In which of the following cases would the resultant force acting on an object be equal to 0N?
- A. A car moving with a constant speed.  
B. A ball falling from the sky to the ground with a constant acceleration.  
C. A person standing on the ground with a constant velocity.  
D. A car moving with a constant acceleration.
- 1.3 Which of the statements below best describes Weightlessness?
- A. An object is not experiencing any forces.  
B. An object is moving at a constant velocity in a straight line.  
C. An object is not experiencing any forces other than the gravity of the Earth.  
D. An object is moving with a constant velocity in a straight line.
- 1.4 The inertia of an object...
- A. is a measure of its mass.  
B. is a measure of its speed.  
C. is a measure of its acceleration.  
D. is a measure of its momentum.
- 1.5 An astronaut has a weight  $F_g$  on Earth. What distance, in terms of the radius of the Earth ( $r$ ), will the astronaut be if his weight is a 1/4 of his weight ( $F_g$ ) on the

surface of the Earth?

- A.  $r$   
B.  $2r$   
C.  $1.5r$   
D.  $0.5r$
- 1.6 A box is pushed forward with a force of 100N but does not move. A possible reason for this is that:
- A. The box is too heavy.  
B. The box is too light.  
C. The box is too big.  
D. The box is too small.

[14]

### QUESTION 2

The grade 11 learners learn that there is more than one way to determine the resultant of two vectors. One way is the component method. They want to use this method to find the resultant of the forces acting on a bucket by the ropes from which the bucket is suspended. The bucket is stationary.

- 2.1 Draw the force diagram. (2)
- 2.2 The tension in rope P is 100N at an angle of  $30^\circ$  to the vertical and the tension in Q is 51,52N at  $45^\circ$  to the vertical.



Calculate the MAGNITUDE of the:

- 2.2.1 Horizontal components of P and Q respectively. (4)  
2.2.2 Vertical components of P and Q respectively. (4)  
2.2.3 Resultant of the horizontal components. (5)  
2.2.4 Resultant of the vertical components. (5)  
2.2.5 Resultant of the forces acting on the bucket. (5)
- 2.3 Calculate the mass of the bucket. (3)

[14]

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**Karin H. Kelder, Jagathesan  
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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web chemistry lab 117 experiment 3 pressure volume and temperature discovery of the gas laws lab partner amazona tahbou october 8 2014 summary the purpose of this experiment is to be able to understand the relationship between the pressure volume and temperature of a gas understanding this relationship is important as it allows

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

*11 5 charles s law volume and temperature - Feb 19 2022*

web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at

constant pressure charles's law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

*9 2 relating pressure volume amount and temperature the* - Mar 03 2023

web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

lab pressure lab reports lab pressure volume and temperature - Oct 10 2023

web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law  $pV = nRT$  is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

8 2 relating pressure volume amount and temperature the - May 05 2023

web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

lab report 5 docx pressure volume and temperature - Aug 28 2022

web pressure volume and temperature discovery of gas laws experiment 5 10 3 19 jared caviglia section 582 ian nicholson introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

*pressure volume temperature lab ku school of engineering* - Jun 25 2022

web the pressure volume temperature pvt fluids lab in ku's chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

**lab report pressure volume and temperature discovery** - Jan 01 2023

web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

*pressure temperature relationship in gases lab studocu* - Nov 30 2022

web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

**pressure volume temperature pvt laboratory department** - Sep 09 2023

web aug 13 2021 pressure volume temperature pvt laboratory analysis of fluid properties such as api gravity viscosity surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and temperature relationships of hydrocarbons are studied at the graduate level

**che144 thermodynamics engineering lab report saturation vapor pressure** - Jul 27 2022

web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

*6 3 relationships among pressure temperature volume and* - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts