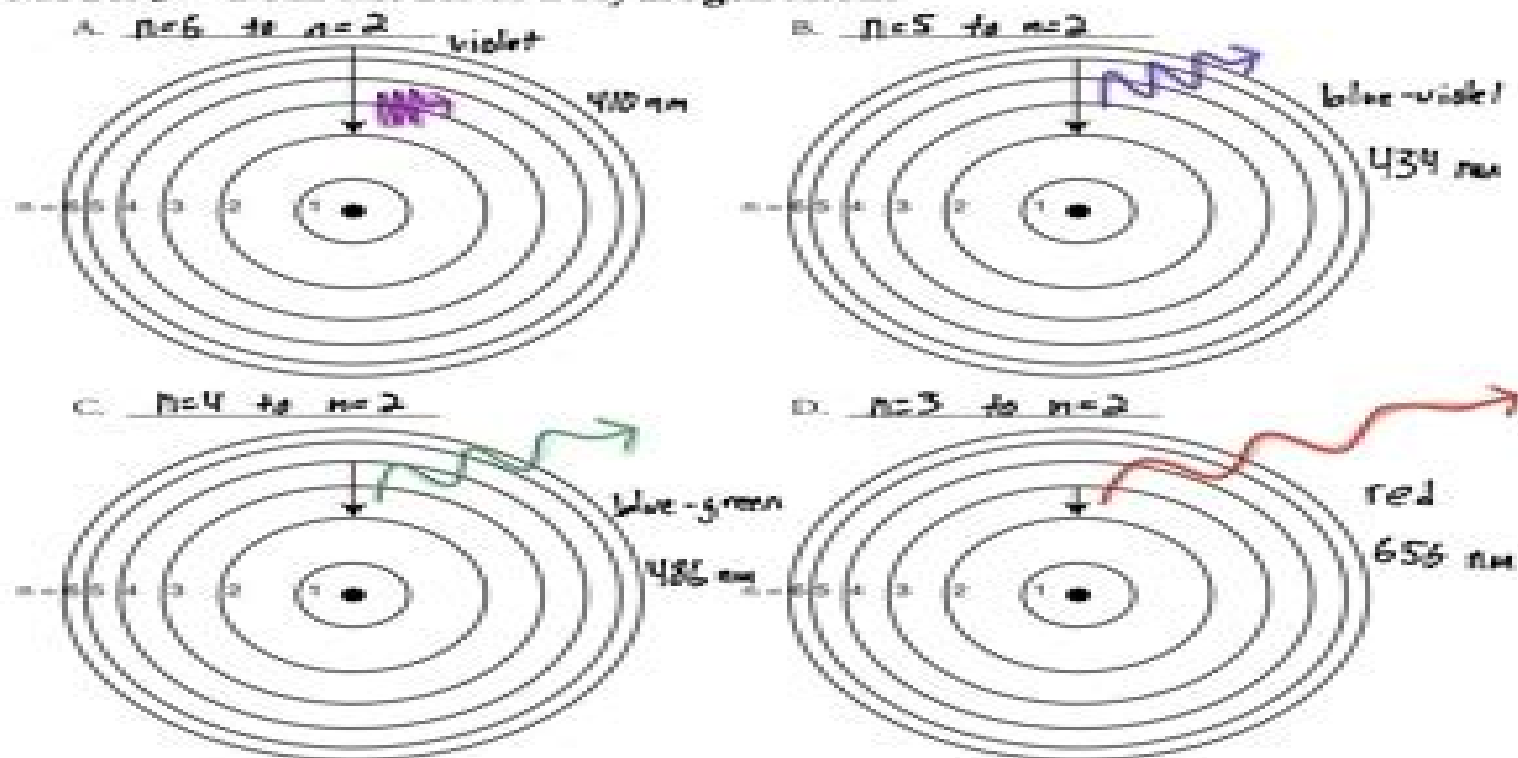


Model 3 – Bohr Model of a Hydrogen Atom



19. Identify the drawing in Model 3 that depicts a hydrogen atom with an electron moving from energy level 5 to energy level 2. Refer to Models 1 and 2 for the following questions.
- Label the picture with " $n=5$ to $n=2$ " and list the corresponding color of light emitted.
 - This electron transition (absorbs/releases) energy.
 - This electron moves from a (lower/higher) energy state to a (lower/higher) energy state.
 - Is light absorbed or released in the electron transition?

Light is released in the electron transition.

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