
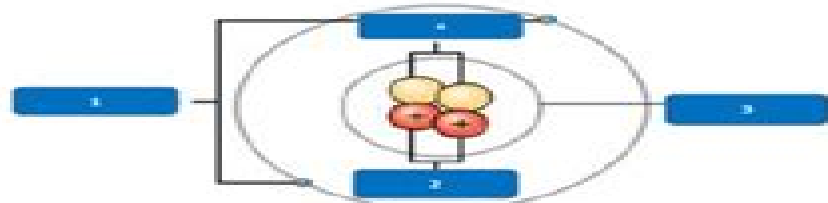
	DAILY LESSON LOG		School Batasan Hills National High School	Grade Level Learning Area	8 Chemistry																				
Teacher MARIA RIZZA E. LLORENTE		Teaching Date and Time for 50 minutes	Quarter 3rd																						
Date: March 23-24, 2023		Day: Thursday-Friday																							
DLL#68																									
I. OBJECTIVE:																									
A. Content Standard		The learners demonstrate understanding of the identity of a substance according to its atomic number																							
B. Performance Standard		The learners should be able to present how water behaves in its different states within the water cycle.																							
C. Learning Competencies		Determine the number of protons, neutrons, and electrons in a particular atom; (S8MT-IIIc-f-10)																							
Specific Objectives:		At the end of the lesson, the students should be able to: 1. Describe the atom and its sub-atomic particles; 2. Determine the number of electrons, protons, and neutrons in an atom; and 3. Explain how atoms benefit life.																							
II. CONTENT		SUBATOMIC PARTICLES Determining the number of protons, electrons and neutrons in an atom																							
III. LEARNING RESOURCES																									
A. References																									
B. Other Learning Resources		<ul style="list-style-type: none">• http://chemed.chem.purdue.edu/genchem/topicreview/bp/ch6/atom_emfame.html• http://www.kentchemistry.com/links/AtomicStructure/protonsws1.htm• https://education2eb.org/lesson_number.html• https://www.nap.edu/read/4818/chapter/2• https://www.youtube.com/watch?v=1M4jysdgmRg																							
IV. PROCEDURE																									
A. Reviewing previous lesson or presenting the new lesson.		ELICIT: TRIVIA QUIZ GAME The class will play trivia quiz game about atoms and molecules  https://www.canva.com/design/DAP6p7UYH0Q8kUQleQA8isLMxgQK1CZueA/edit																							
B. Establishing a purpose for the lesson		ENGAGE: LABEL THE PICTURE: What's Inside the Atom?  a. Ask the class to label the picture. b. Introduce subatomic particles. c. Recall the proponent/discovery of the atomic model on the picture. Answer: 1. Electron 2. Proton 3. Nucleus 4. neutron																							
C. Presenting examples/instances of the new lesson		ENGAGE: TABLE ANALYSIS The students will study the table below and answer the questions about Properties of Subatomic Particles. <table border="1"><thead><tr><th>Subatomic particle</th><th>Symbol</th><th>Relative charge</th><th>Location</th><th>Mass</th></tr></thead><tbody><tr><td>Proton</td><td>p</td><td>+1</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Neutron</td><td>n</td><td>0</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Electron</td><td>e⁻</td><td>-1</td><td>Outside the nucleus</td><td>0.0005 (-1/1836)</td></tr></tbody></table>				Subatomic particle	Symbol	Relative charge	Location	Mass	Proton	p	+1	Inside the nucleus	1	Neutron	n	0	Inside the nucleus	1	Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)
Subatomic particle	Symbol	Relative charge	Location	Mass																					
Proton	p	+1	Inside the nucleus	1																					
Neutron	n	0	Inside the nucleus	1																					
Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)																					

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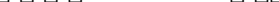

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