Same \_\_\_\_\_

# Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- a. Write the balanced chemical equation
- b. Identify the given (with units) and what you want to find (with units)
- c. Show set up with units. Check sig figs, give final answer with units and label.
- Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?

Equation:	2 H <sub>2</sub> O <sub>(i)</sub> →	2 H2(c) +	1 O2 (c)
Before	2 mol	0 mod	0 mod
Change	-2 mol	+2 mol	
After	0 mol	2 mol	

$$36 \text{ g } H_2Ox \frac{1 \text{ mol } H_2O}{18.0 \text{ g } H_2O} = 2.0 \text{ mol } H_2O \times \frac{2 \text{ mol } H_2}{2 \text{ mol } H_2O} = 2 \text{ mol } H_2 \times \frac{2.0 \text{ g } H_2}{1 \text{ mol } H_2} = 4.0 \text{ g } H_2$$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?

Equation:	2 Na <sub>(0)</sub> +	$-$ L $Cl_{2(g)} \rightarrow$		-
Before	9.94 mol	xs mol	0 mod	
Change	-9.94 mod		+9.94 mol	
After	0 mod		9.94 mol	

$$581 \ g \ NaClx \\ \frac{1 \ mol \ NaCl}{58.5 g \ NaCl} = 9.93 \ mol \ NaCl \\ \times \\ \frac{2 \ mol \ Na}{2 \ mol \ NaCl} = 9.93 \ mol \ Na \\ \times \\ \frac{23.0 \ g \ Na}{1 \ mol \ Na} = 228 \ g \ Na \\ \frac{1 \ mol \ Na}{2 \ mol \ Na} = 228 \ g \ Na \\ \times \\ \frac{1 \ mol \ Na}{2 \ mol \ N$$

3. You eat 180.0 g of glucose (90 M&Ms). If glucose, C6H12O6, reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to breathe in to burn the glucose?

Equation:	1 CeH <sub>12</sub> O <sub>6(e)</sub> +	6 O₂(g) →	6 CO <sub>2(g)</sub>	+ 6 H <sub>2</sub> O <sub>(g)</sub>
Before	1.0 mod	xs mol	0 mod	0 mol
Change	-1.0 mol	-1.0 mol		- 33004
After	0 mol	xs mol		

$$180.0 \text{ g C}_{a}H_{12}O_{a}x \frac{1 \text{ mol C}_{a}H_{12}O_{a}}{180 \text{ g C}_{a}H_{12}O_{a}} = 1.00 \text{ mol C}_{a}H_{12}O_{a} \times \frac{6 \text{ mol } O_{2}}{1 \text{ mol C}_{a}H_{12}O_{a}} = 6.00 \text{ mol } O_{2}$$

6.0 mol 
$$O_2 x \frac{32.0 \text{ g O}_2}{1 \text{ mol O}_2} = 192.0 \text{ g } O_2$$

# **Lesson 28 Stoichiometry 1 Answer Key**

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General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and guestions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 guestion practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material MCAT Workout, 2nd Edition Princeton Review, 2018-12-18 Make sure you re studying with the most up to date prep materials Look for The Princeton Review s MCAT Workout Revised 3rd Edition ISBN 9780525570080 on sale October 2019 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original AP Chemistry Premium, 2022-2023: Comprehensive Review with 6 Practice Tests + an Online Timed **Test Option** Neil D. Jespersen, Pamela Kerrigan, 2021-07-06 Be prepared for exam day with Barron's Trusted content from AP experts Barron's AP Chemistry Premium 2022 2023 includes in depth content review and online practice It's the only book you ll need to be prepared for exam day Written by Experienced Educators Learn from Barron's all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent exam Get a leg up with tips strategies and study advice for exam day it s like having a trusted tutor by your side Be Confident on Exam Day Sharpen your test taking skills with 6 full length practice tests 3 in the book and 3 more online Strengthen your knowledge with in depth review covering all Units on the AP Chemistry Exam Reinforce your learning with practice questions at the end of each chapter Interactive Online Practice Continue your practice with 3 full length practice tests on Barron's Online Learning Hub Simulate the exam experience with a timed test option Deepen your understanding with detailed answer explanations and expert advice Gain confidence with automated scoring to check your learning progress 2019-2020 Kaplan Test Prep, 2019-01-01 Kaplan's DAT Prep Plus 2019 2020 provides the test taking strategies realistic practice and expert guidance you need to score higher on the Dental Admissions Test Our comprehensive updated subject review reflects recent changes to the blueprint of the exam question types and test interface You ll get two full length practice DATs and expert tips to help you face Test Day with confidence The Best Review Two updated full length online practice exams for test like practice Study planning guidance More than 600 practice questions for every subject with detailed answers and explanations Full color study sheets for high yield review A guide to the current DAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the DAT Expert Guidance Our books and practice questions are written by veteran teachers who know students every explanation is written to help you learn Kaplan's experts ensure our practice questions and study materials are true to the test We invented test prep Kaplan

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