PRACTICE TEST-1 FEB-01-14 - Practice Test for CBSE-2014 Board Exams

Time allowed: 180 min

Marks: 100

General Instructions

- The question paper consists of 3 parts. Each question of each part is compulsory.
- SEC-A: (10 questions) Questions 1 to 10 are of 1 mark(s) each
- SEC-B: (12 questions) Questions 11 to 22 are of 4 mark(s) each
- SEC-C: (7 questions) Questions 23 to 29 are of 6 mark(s) each
- Links to individual answers on clay6.com have also been provided.

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$$(10 \times 1 \text{ mark} = 10 \text{ marks})$$

- 1) What is the angle between following planes 2x-3y+4z=1 and -x+y=4?
- 2) What is the area of the parallelogram having diagonals $3\vec{i} + \vec{j} 2\vec{k}$ and $\vec{i} 3\vec{j} + 4\vec{k}$?
- 3) If $e^{x+y} = xy$, what is $\frac{dy}{dx}$?
- Given $A=egin{bmatrix} 3&-3&4\\2&-3&4\\0&-1&1 \end{bmatrix}$ Express A^{-1} in terms of A?
- **5)** What does $\cos^{-1} \frac{4}{5} + \cos^{-1} \frac{12}{13}$ reduce to?
- If $\begin{bmatrix} xy & 4 \\ z+6 & x+y \end{bmatrix} = \begin{bmatrix} 8 & w \\ 0 & 6 \end{bmatrix}$ then find the value of x,y,z and w
- 7) If A,B and C are angles of a triangle , then the determinant $\begin{vmatrix} 1 & \cos C & \cos B \\ \cos C & 1 & \cos A \end{vmatrix}$ is equal to $\cos B \cos A = 1$
- 8) Integrate the function $e^x \left(\frac{1}{x} \frac{1}{x^2} \right)$
- 9) Let $f: \{1,3,4\} \to \{1,2,5\}$ and $g: \{1,2,5\} \to \{1,3\}$ be given by $f = \{(1,2),(3,5),(4,1)\}$ and $g = \{(1,3),(2,3),(5,1)\}$. Write down gof.
- 10) Choose the correct answer. The planes: 2x y + 4z = 5 and 5x 2.5y + 10z = 6 are:

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Chao Zhang

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In a world driven by information and connectivity, the ability of words has be much more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Grade 12 Maths Question Paper 2014**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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web at very high pressures a gas will occupy a larger volume than predicted by the ideal gas law which of the following statements is true at low temperatures intermolecular forces become important and the pressure of a gas will be lower than predicted by