

12. For the first five hours, the concentration of  $F_2$  decreased from 0.730 mol to 0.434 mol.



What is the reaction rate of  $F_2$ ? The unit that must be used is  $M/h$ . Express your answer using scientific notation.

13. For Question 12, suppose that  $O_2$  is reacting at the rate of 2.00  $M/h$  for three hours. It is observed that 5.00 mol has been reduced. What is the initial concentration? Round-off your answer to four decimal places.

14. Consider the following reaction:



Scientists who are working at a certain laboratory institution found out that the concentration of  $H_2O_2$  is decreasing with time. They noted that its concentration before the reaction began was 1.2 mol. After the time elapsed, they observed that 0.57 mol has been decreased from the original amount. The reaction rate of  $H_2O_2$  is 3.35  $M/h$ . How much time does it take for this to occur? The unit of your answer must be the SI unit of time.

# Lesson 36 Reaction Rates Answer Key

**R Barnett**



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