

29/9/2023

Ionic Bonding - Unit 1 - Aim A - Lesson 4

Ionic bonding - the attraction of ions that are positive and negative

Common Anions and Cations -

Cation - Hydrogen Ion - H^+

Anion - Hydride - H^-

Cation - atom or molecule that is negatively charged

Anion - an atom or molecule that is positively charged

5.0g of F_2 - 0.13

1. $2(19) = 38$

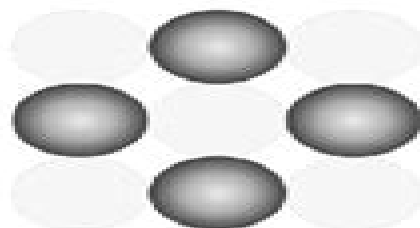
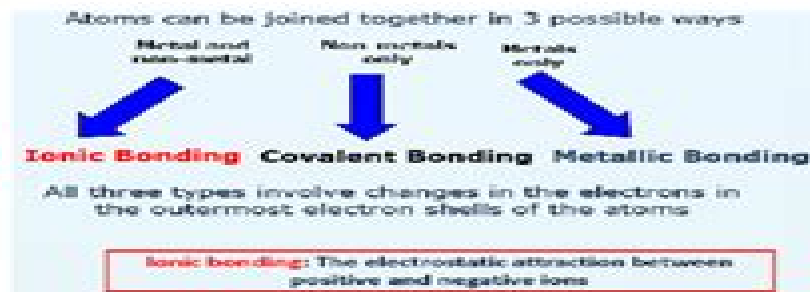
2. $5/38 = 0.13$

34.5g of NO_2 - 0.75

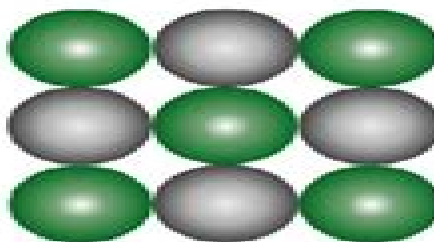
1. $14 + 2(16) = 46$

2. $34.5/46 = 0.75$

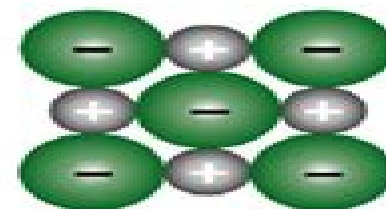
Moles = mass/M_r (relative formula mass)



Sodium chloride contains sodium ions.



Sodium chloride contains sodium ions.



Sodium chloride contains negatively charged ions.

Lesson 20 Covalent Bonds Unit 4

D Keegan



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