

7 Ionic and Metallic Bonding

Big idea BONDING AND INTERACTIONS

7.1 Ions

For students using the Foundation edition, assign problems 1, 3–5, 7–12, 14, 15, 18–20.

Essential Understanding Ions form when atoms gain or lose valence electrons, becoming electrically charged.

Lesson Summary

Valence Electrons Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation.

- ▶ For a representative element, the group number equals the number of valence electrons the atom contains.
- ▶ An electron dot structure shows the symbol of the element and its valence electrons.
- ▶ Atoms tend to gain or lose the number of electrons that will provide the atom with a noble gas electron configuration.

Formation of Cations Cations are positively charged ions formed when an atom loses one or more valence electrons.

- ▶ Atoms and the cations formed from them have different properties.
- ▶ Elements in Group 1 form cations with a charge of $1+$, and those in Group 2 form cations with a charge of $2+$.
- ▶ Many transition metals form more than one cation and do not follow the octet rule.

Formation of Anions Anions are negatively charged ions formed when an atom gains one or more valence electrons.

- ▶ Commonly, the name of an anion ends in *-ide*.
- ▶ Anions form from nonmetallic elements.
- ▶ The anions formed from halogens are known as halides.

After reading Lesson 7.1, answer the following questions.

Valence Electrons

1. What are valence electrons?

Valence electrons are the electrons in the highest occupied energy level of an element's atoms.

2. The valence electrons largely determine the *chemical properties* of an element and are usually the only electrons used in *chemical bonds*.

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Reading & Writing Japanese: A Workbook for Self-Study Eriko Sato, 2021-10-12 The easy way to learn the Japanese writing system In this beginner's workbook students of Japanese will learn to read and write the Hiragana and Katakana alphabets as well as 50 basic Kanji characters A step by step approach moves from tracing letters to reading and writing complete words quickly and effectively In this comprehensive guide language learners will find The 92 Hiragana and Katakana letters 50 Kanji characters Illustrations to help you remember each letter and its associated sound An extensive bidirectional dictionary Hundreds of drills suitable for self study or classroom use An answer key for all exercises Useful everyday vocabulary and phrases from numbers and colors to food and social media Free printable flashcards to help you review what you've learned Online audio recordings by a native speaker to help with pronunciation Instead of dealing with separate Hiragana and Katakana workbooks as is common this comprehensive guide provides an efficient way to learn both with the help of one book

Emergency medical services United States. National Highway Traffic Safety Administration, 1979

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