Unit 9: Heat

Things to be Studied in this Unit

- Thermal energy, heat and temperature
- Anomalous expansion of water and its effect
- Effect of heat on volume
- Thermometer and its structure
- Thermometric liquids (Mercury and Alcohol)
- Types of thermometers and their uses
- Specific heat capacity
- Heat equation
- Simple numerical problems related to heat

Introduction

Heat is a form of energy that is transferred between two substances due to a temperature difference. It flows spontaneously from a hotter object to a colder one. Heat is measured in calories or joules. Heat is associated with the kinetic energy of atoms and molecules within a substance. It is what causes substances to change in temperature.

Temperature is a measure of the average kinetic energy of the particles (atoms or molecules) within a substance. It is measured in degrees Celsius (°C) or degrees Fahrenheit (°F) and kelvin (K).

Things to Know

- Heat energy: Heat energy of an object is the sum of the kinetic energy contained by the molecules of that object.
- One calorie heat energy: One calorie of heat energy is the quantity of heat required to raise the temperature of one gram of pure water by 1°C or 1K.
- One joule heat energy: One joule of heat energy can be defined as the amount of heat energy required to raise the temperature of an object of mass 1 kg and specific heat capacity of 1 Jkg⁻¹°C⁻¹ by 1°C.
- Anomalous expansion of water: Generally, liquids expand on heating and contract on cooling. But a different behaviour is observed in the case of water. When water at 0°C is heated, it contracts until the temperature rises to 4°C. But expansion is observed after 4°C only. This unusual behaviour shown by water between 0°C and 4°C is called the anomalous expansion of water.

At 4°C, water has maximum density and minimum volume. This property of water is beneficial for aquatic animals in cold countries.

Advantages of anomalous expansion of water:

- In cold countries, fishes and other aquatic animals can survive in the ponds even though the surface has frozen to ice.
- In cold countries, soft drink bottles can be stored below the frozen ice of the pond to prevent them from freezing.

Disadvantages of anomalous expansion of water:

- Water pipes burst in cold countries due to expansion of water inside the pipe when the temperature of water falls below 4°C to 0°C.
- 2. Fruits and vegetables get damaged during severe frost in cold countries.

Hs Science Unit 9 Lesson Answers

Wilbur Lee Beauchamp

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Teaching High School Science Through Inquiry Douglas Llewellyn, 2005 This is the secondary school I version of Llewellyn's strong Corwin debut Inquire Within Implementing Inquiry Based Science Standards 2000 This book focuses on raising a teacher's capacity to teach science through an inquiry based process implementing inquiry as stated by the national Teaching High School Science Through Inquiry and Argumentation Douglas Llewellyn, 2013 For Grades 9 12 standards this new edition covers assessment questioning techniques to promote learning new approaches to traditional labs and activities that emphasize making claims and citing evidence Resources in Education ,2001 The High School Teacher ,1925 **Research in Education** ,1973 Course of Study in Physical Science for the Junior High School, Grades 7, 8, and 9 New Jersey. Department of Public Instruction, 1927 United States History Mark C. Schug, National Council on Economic Education, 1993 Supplementing secondary U S history textbooks this blends historical facts and economic reasoning through case studies lectures and class discussion Mathematics and Science Across the Curriculum ,2002 GED Testing United States. Indian Health Service, 1973 **Programmed Learning and Individually Paced** Preparation for High School Equivalency in Science Philip J. Gearing, Craig **Instruction** Carl H. Hendershot.1973 Conrad,1978 Instructional Strategies for Middle and High School Social Studies Bruce E. Larson, 2016-08-10 Instructional Strategies for Middle and High School Social Studies Methods Assessment and Classroom Management is an exciting methods based text that integrates appropriate management and assessment techniques with seven distinct teaching strategies Writing explicitly for pre service social studies teachers veteran teacher educator Bruce E Larson offers detailed descriptions of a range of instructional strategies along with guidelines for deciding how and when to use each Part I offers the foundations for teaching and learning in a social studies classroom and explores contextual theoretical and policy factors that all teachers need to consider before entering the classroom Part II delivers a range of comprehensive strategies for providing instruction that is appropriate for particular lessons student abilities and classroom environments The practical strategies in Part II build upon the learning theories described in Part I positioning Instructional Strategies for Middle and High School Social Studies to be the go to all inclusive teacher's guide to the social studies classroom New to this Edition A list of goals before each chapter presents an overview of the chapter's content focus and provides an outline for the chapter review Extensively revised Part I chapters 1 4 provides an updated review of national standards developed for teaching history geography civics and economics In depth applications of the Common Core State Standards for the social studies are also explored New Reality Check feature provides directions for integrating field based experiences into the chapters and contextualizes the ideas in the book for a classroom setting Each chapter in Part II chapters 5 11 has been expanded to include a section labeled Enhancing Student Learning with Technology offering websites links and other resources for integrating recent technologies into the classroom Chapters 5 11 include a new Making Your Lesson More Meaningful for

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