

Guidelines for Writing a Formal Laboratory Report

Scientific writing shares with all expository writing the goal of conveying information as clearly as possible to the reader. Chemists have developed a set of conventions to make the results of their work easily accessible to their peers; by following the same conventions for your lab reports, you allow other chemists to analyze your work. The following outlines briefly how to apply these conventions to your formal laboratory reports in chemistry courses at Smith. Your lab instructor may provide examples and/or additional instructions specific to your current course. Other valuable sources include: chemistry journals (e.g. *Journal of the American Chemical Society*), the *ACS Style Guide* (available in Young Science Library or from your instructor) and the Jacobson Center.

Structure of lab reports

Reports will be divided generally into the following sections, in order of appearance:

1. **Abstract (optional):** A short (~ 100 words) description of what was done and the major result(s) and conclusion(s).
2. **Introduction:** This section tells the reader why you did the experiment. It includes some or all of the following: background information, an explanation of the hypothesis you set out to test and the results of previous, closely-related experiments. When the reader finishes reading the introduction, she should know what to expect in the report.
3. **Experimental:** This section (called "Materials and Methods" in biology) describes what you did in enough detail that someone else with your level of experience could repeat the experiment. Your description must therefore be quantitative (for instance, what temperature did you heat the mixture to and for how long?) as well as specific (for instance, what are the make and model of the spectrometer you used for analysis?). Note that this section sometimes comes at the end of the lab report (particularly in organic chemistry); your instructor will tell you if that is the case.
4. **Results:** In this section you present the data from your experiment. (Note that some simple data, such as the yield from a synthetic step, may be presented in the Experimental section; check with your instructor for your specific lab.) You may use tables and/or figures to present the results, but you should describe any relevant features of the results completely within the text, referring the reader to the appropriate table or figure as necessary. Keep in mind that tables are useful when the reader wants to know the exact numerical value of a result, while graphs are useful for showing trends and for deriving values from a statistical fit of data. Both tables and figures should be numbered sequentially, and each should have a descriptive title. **Tables** may be presented in any format that clearly delineates rows and columns from each other. Headings should indicate the units associated with any value. **Figures** may include any graphic information such as schematics of experimental apparatus and drawings of large molecules; in your lab reports figures will most often be graphs. The same instructions apply to graphs as to the rest of your report: present the material in the simplest and clearest manner possible. Your instructor may give you further hints.

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E Durkheim



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