

# Experiment Problems 1

1. Synthesis reactions:



2. Synthesis by mixing method in dilute solutions:



3. Synthesis solubility product (K<sub>sp</sub>) constants and equilibrium hydrolysis and acids:

|    | Elemental        | Basic Difference  |
|----|------------------|---|
| a) | $\text{Pb}^{2+}$ |   |
| b) |                  | $2\text{H}^+ + \text{CO}_3^{2-} \rightarrow \text{H}_2\text{O} + \text{CO}_2$ |
| c) | $\text{Ca}^{2+}$ |   |

|    | Elemental        | Basic Difference             |
|----|------------------|------------------------------|
| a) | $\text{Pb}^{2+}$ |                              |
| b) | $\text{Ca}^{2+}$ |                              |
| c) |                  | $\frac{1}{\text{CO}_3^{2-}}$ |

4. Synthesis equilibrium constants: chemical constants are represented hydrolysis and acids:



# Mathematics Unit 1 Lesson 01 Tesccc Answers

**Liying Dong**



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