- 2 -

- Which of the following processes provide evidence of the particulate nature of matter?
 - I. Capillarity
 - II. Diffusion
 - III. Filtration
 - IV. Osmosis
 - (A) I and III only
 - (B) I and IV only
 - (C) II and III only
 - (D) II and IV only
- 'Mass number' is the number of
 - (A) neutrons plus protons
 - (B) neutrons minus protons
 - (C) electrons plus neutrons
 - (D) electrons plus protons
- 3. Which of the following elements has 7 electrons in its outer shell?
 - (A) Hydrogen
 - (B) Oxygen
 - (C) Nitrogen
 - (D) Chlorine
- An atom that has 20 electrons when it ionizes has the same electronic configuration as
 - (A) argon
 - (B) calcium
 - (C) neon
 - (D) sodium
- ³⁴Na and ²⁵Na are isotopes because they
 - (A) conduct electricity
 - (B) react vigorously with water
 - C) have the same number of protons
 - (D) have the same relative atomic mass

- Sulfur and oxygen are in the same group of the periodic table because
 - (A) they can react with each other
 - (B) they can form covalent compounds
 - (C) they have the same number of electrons in their outer shell
 - (D) the atomic number of sulfur is 16 and the relative atomic mass of oxygen is 16

<u>Item 7</u> refers to the following quantities of atoms.

- 12 g of carbon
- II. 16 g of sulfur
- III. 23 g of sodium
- IV. 12 g of magnesium

(Relative mass number: C = 12; S = 32; Mg = 24; Na = 23)

- 7. Which of the following pairs represents equal quantities of atoms?
 - (A) I and II only
 - (B) I and IV only
 - (C) II and III only
 - (D) II and IV only
- An ion with a single negative charge may be converted into a neutral atom by
 - (A) losing a neutron
 - (B) gaining a proton
 - (C) losing an electron
 - (D) gaining an electron

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- 1. Understanding the eBook Gcse Physics June 2015 Answers
 - The Rise of Digital Reading Gose Physics June 2015 Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Gcse Physics June 2015 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Physics June 2015 Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Gcse Physics June 2015 Answers
 - Personalized Recommendations
 - Gcse Physics June 2015 Answers User Reviews and Ratings
 - $\circ\,$ Gcse Physics June 2015 Answers and Bestseller Lists
- 5. Accessing Gose Physics June 2015 Answers Free and Paid eBooks
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 - Gcse Physics June 2015 Answers eBook Subscription Services

- Gcse Physics June 2015 Answers Budget-Friendly Options
- 6. Navigating Gcse Physics June 2015 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gcse Physics June 2015 Answers Compatibility with Devices
 - Gcse Physics June 2015 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcse Physics June 2015 Answers
 - Highlighting and Note-Taking Gose Physics June 2015 Answers
 - Interactive Elements Gcse Physics June 2015 Answers
- 8. Staying Engaged with Gcse Physics June 2015 Answers
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gose Physics June 2015 Answers
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Physics June 2015 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gose Physics June 2015 Answers
 - Setting Reading Goals Gose Physics June 2015 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gcse Physics June 2015 Answers
 - Fact-Checking eBook Content of Gcse Physics June 2015 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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